

3.14

$^{12}\text{C}$  atom has mass  $12\text{u}$  exactly  
 the nucleus = atom - 6 (electrons)  
 mass of 6 electrons =  $6 (0.549 \times 10^{-3}\text{u})$   
 $= 3.294 \times 10^{-3}\text{u}$

So mass of  $^{12}\text{C}$  nucleus =  $(12.000 - 0.003)$   
 $= 11.997\text{u}$

3.23

1 pound mass = 454 grams

12 grams = 1 mole of  $^{12}\text{C}$

1 mole of  $^{12}\text{C} = (N_A)(12)\text{u}$

So  $N_A\text{u} = 1\text{gram}$

part b)  $6.022 \times 10^{23}\text{u} = 1\text{gram}$

a.) mult. by 454:  $2.73 \times 10^{26}\text{u} = 1\text{pound mass}$

3.32

$$\lambda = \frac{l}{n\sigma} ; \quad pV = NkT$$

$$p = \frac{N}{V} kT = nkT$$

so  $\lambda = \frac{kT}{p\sigma} = d$  for Knudsen regime

$p = \frac{kT}{d\sigma}$  if we use S.I. units

$$p = \frac{1.38 \times 10^{-23} (293)}{(20 \times 10^{-2}) \times (0.15 \times 10^{-9})^2} = 0.29 \text{ pascal}$$

one atm  $\approx 10^5$  pascal

so  $p = 0.29 \times 10^{-5} = 2.9 \times 10^{-6} \text{ atm}$

3.37

- eq 3.36  $\rightarrow D_{rms} = \left( \frac{kT}{3\pi^2 a} \right)^{1/2} t^{1/2}$

the only thing changed is a

3.37 - cont'd

Since mass does not appear here

$Q$  is doubled, so  $D_{rms}$  is multiplied by  $\frac{1}{\sqrt{2}} \rightarrow D_{rms} = 71 \mu m$

3.38

here we are asked how far it goes if  $t \rightarrow 10 \times t$ .

Since  $D_{rms}$  is proportional to  $\sqrt{t}$   
 $D_{rms}$  will be multiplied by  $\sqrt{10}$   
 $= 10 \times \sqrt{10} \mu m = 32 \mu m$

3.40

$$\vec{\Delta p} = \vec{F} \Delta t$$

$$\vec{p}_0 = m v \hat{x}$$

$$\vec{F} = -e E \hat{y}$$

$$\Delta t = l/v$$

$$\vec{p}_0$$

$$\vec{p}_F$$

$$\tan \theta = -p_y/p_x$$

$$p_y = \Delta p$$

since  $\vec{F}$  is in  $\hat{y}$  direction and  $p_y$  started at 0

$$p_x = p_0$$

since  $F_x = 0$

$$\text{So } \tan \theta = \frac{e E l / v}{m v} = \frac{e E l}{m v^2}$$

and  $\tan \theta \approx \theta$  for small  $\theta$ .