

Key

NAME:

Show all your work. You are welcome to use a calculator but no notes, books, internet resources (Desmos is the exception!) or peers can be used. Reasonable supporting work must be shown to earn credit.

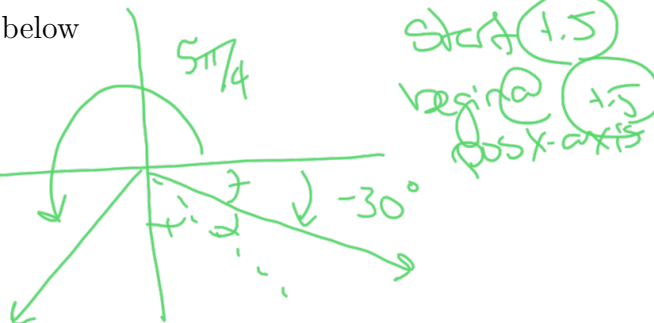
1. Sketch the angles with the measures given below

(a) [2] -30°

direction (+.5)
space of angle (+.5)
terminal side (+.5)

(b) [2] $\frac{5\pi}{4}$ radians

direction (+.5)
space of angle (+.5)
terminal side (+.5)



2. Provide a graph AND an algebraic rule/expression that which is described:

(a) [3] A exponential function whose range (outputs) is $(-2, \infty)$.

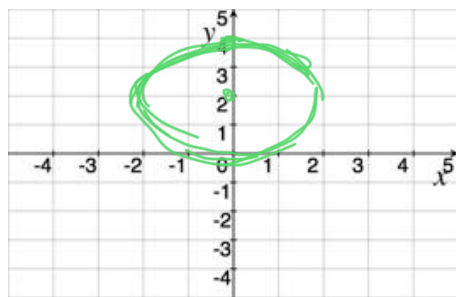


$y = 2^x - 2$ works

exponential (+.5) match (+.5)
shifts (+.5) start (+.5)

+ .5 exp shape
+ .5 range

(b) [3] A circle with radius 2 and centered at $(0, 2)$



Note: looks like an oval b/c the scales on y axis are dif. than those on x-axis

$(y-2)^2 + (x-0)^2 = 2^2$

eq of circle (+.5)
center (+.5) match (+.5)
radius (+.5)

center/radius (+.5)
shape (+.5)

Activity 67
Webwork #6

§ 3.2 #56
and webwork #8

3. Let f be a logarithm function (whose base is unknown!!!) that has been horizontally shifted & graphed below.

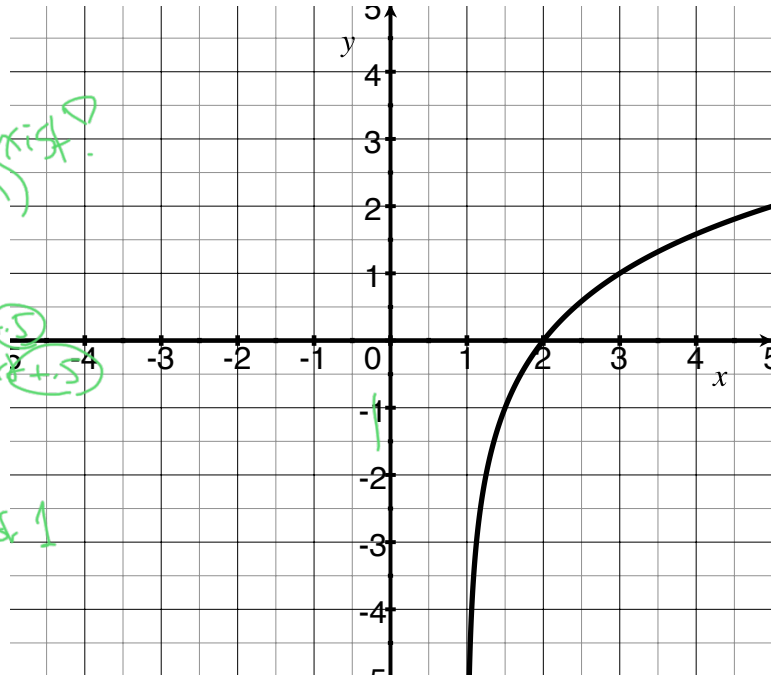
(a) Estimate the following if possible:

i. [1] $f(0)$ Does not exist.
(trick question)
(+1)

ii. [2] the domain of f
 $(1, \infty)$ int (+.5)
x-values not inc (+.5)
start (+.5)

iii. [1] all a so that $f(a) = 1$.
inputs w/ output of 1
 $a = 3$ (+1)

iv. [1] the x -intercept(s)
 $x = 2$ (+1)



(b) [2] Does f have an inverse? Why or why not?

yes?
(+1.5)

(c) [2] We know from above that the graph of f is shifted horizontally. Describe precisely how much and in what direction $f(x)$ can be obtained from the graph of a basic logarithmic function (whose base is unknown!!!).

start (+.5) $\log_b(x) = y$ passes thru $(1, 0)$ b/c } So the log graph has been shifted right by 1 unit (+.5)
horizontal shift (+.5) $\log_b(1) = 0$

(d) [3] Find the algebraic rule for the function f .

$y = \log_b(x-h)$ logarithmic equation (+.5)
horizontal shift (+.5)

graphs thru $(2, 0)$ (+.5)
 $0 = \log_b(2-h)$ OR horizontal shift right by 1 unit
 $\Rightarrow b^0 = 2-h$
 $\Rightarrow 1 = 2-h$
 $\Rightarrow h+1 = 2$
 $h = 1$ (+.5)
 $\Rightarrow y = \log_b(x-1)$ 2

(+.5) graph goes thru $(3, 1)$
so $1 = \log_b(3-1)$ (+.5)
 $\Rightarrow b^1 = 3-1$ exponential (+.5)
 $\Rightarrow b = 2$ (+.5)
so $y = \log_2(x-1)$

Worksheet #20
Practice Exam #9

Worksheet #21
Quiz #3

4. The temperature T (in $^{\circ}\text{C}$) of coffee at time t minutes after its removal from the microwave is given by the equation $T = 25 + 73e^{-0.28t}$.

(a) [2] Find the temperature when after a half hour has passed.

half an hour = 30 min = t so $25 + 73e^{-0.28(30)} \approx 25^{\circ}\text{C}$

(b) [3] When will the temperature reach 30°C ?

find t when $30 = 25 + 73e^{-0.28t}$

$5 = 73e^{-0.28t}$

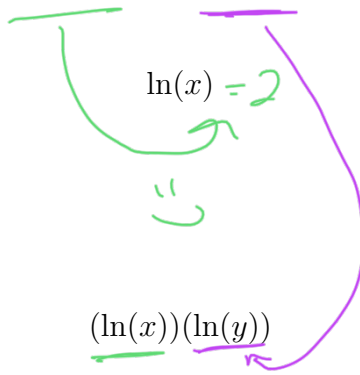
$\frac{5}{73} = e^{-0.28t}$

$\ln\left(\frac{5}{73}\right) = -0.28t$

$t = \frac{-1}{0.28} \ln\left(\frac{5}{73}\right) \approx 9.6 \text{ min}$

5. Let x and y be defined so that $\ln(x) = 2$ and $\ln(y) = 5$. Compute the following:

(a) [1]



(b) [1]

$(\ln(x))(\ln(y))$

$2 \cdot 5$

$= 10$

(c) [2]

+1 property
+1 plugin

$\ln(xy) = \ln(x) + \ln(y)$

$= 2 + 5$

$= 7$

(d) [2]

since $\ln(x) = 2$

$\Rightarrow e^{\ln(x)} = e^2$

$x = e^2$

$x \approx 7.39$

use exp +1.5
use right +1.5

Activity 1/4/5
Worksheet 5

5.3.3 1/6

Word Problems 2

6. Entropy S is a function of the number of possible states W , that are accessible to a system with a given amount of energy. We can explicitly compute entropy by

$$S = k \ln(W)$$

where k is Boltzmann's constant which is approximately $1.38065 \cdot 10^{-23} \text{m}^2 \text{kg s}^{-2} \text{K}^{-1}$.

(a) [3] If a gas has entropy 2, about how many possible states does the gas have?

start (1.5) \downarrow (1.5) S want to find w (4.5)

$$2 = \frac{k \ln(w)}{k}$$

$$2/k = \ln w$$

$$e^{2/k} = w \quad (+1)$$

$w \approx \text{BIG Number?}$

(b) [4] If liquid A has 1,000,000,000,000 (so $1 \cdot 10^{12}$) times more possible states than liquid B, which liquid has a higher entropy and what is the difference?

let w_A be # of states of liquid A define variables (+1)
 let w_B be # of states of liquid B

$w_A = 1 \cdot 10^{12} w_B$ (+1.5)

entropy for A entropy for B

$$k \ln(w_A)$$

$$k \ln(w_B)$$

$$= k \ln(1 \cdot 10^{12} \cdot w_B)$$

$$= k [\ln(10^{12}) + \ln w_B]$$

$$= k \ln(10^{12}) + k \ln w_B$$

$$= k \ln(10^{12}) + \text{entropy of B}$$

(+1) algebra/log prop?

(+1) find relation between entropy of A & B by relating entropy of B

↑ higher entropy by difference (+1.5)

$$k \cdot \ln(10^{12}) = k \cdot 12 \ln(10)$$

$$= 1.38065 \cdot 10^{-23} \cdot 12 \cdot \ln(10)$$

4 \approx small #