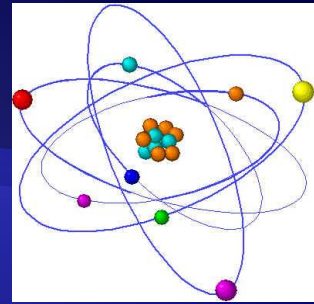


Session 2

# BUILDING BLOCKS

# Atoms



- ◆ Six basic elements

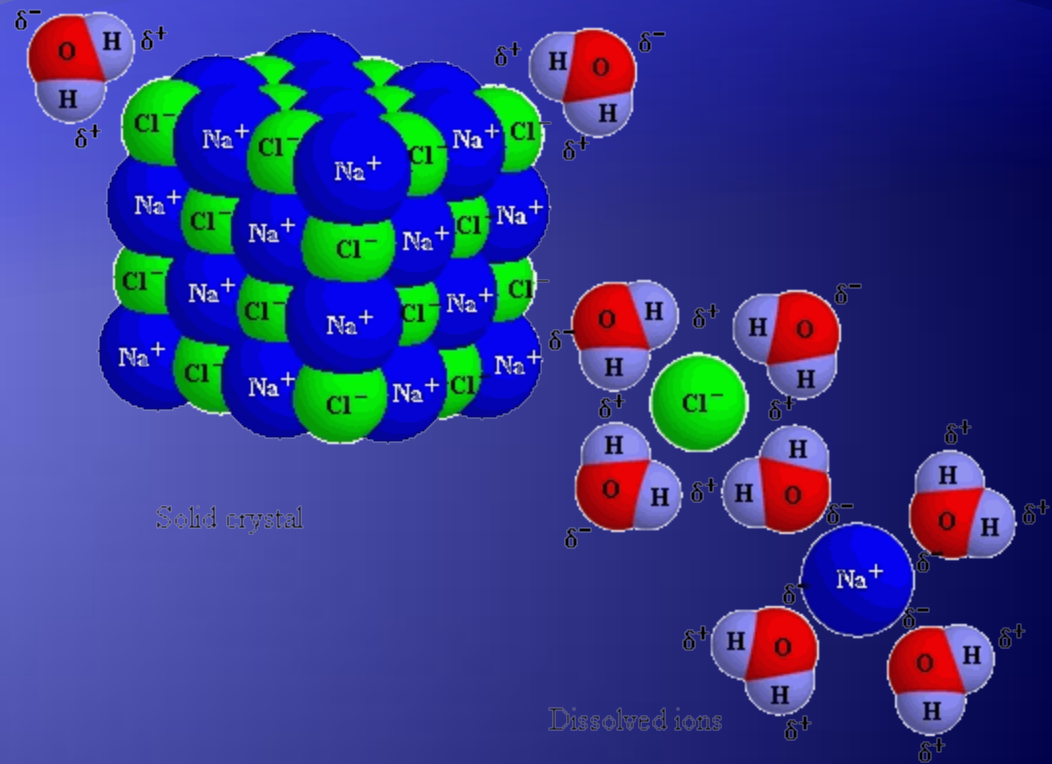
- ◆ Hydrogen (H), Carbon (C), Nitrogen (N), Oxygen (O), Phosphorous (P), Sulfur (S)

Atoms	H	C	N	O	P	S
Valency	1	4	5	6	5	6
Electronegativity	2.1	2.5	3.0	3.5	2.1	2.5
<i>H. Sapien (%)</i>	9.3	19.4	0.8	62.8	0.6	0.6
<i>Bacteria (%)</i>	9.9	12.14	3.0	73.7	0.6	0.3
<i>Earth (%)</i>	0.9	0.2	0.9	50	0.12	0.12

- ◆ Valence electrons determine the bonding
- ◆ Electronegativity determines the polarity
- ◆ Composition is similar between living organisms

# Molecular Bonds

- ♦ Ionic bonds
- ♦ Covalent bonds
- ♦ Hydrogen bonds
- ♦ Van der Waal bonds
- ♦ Hydrophobic interactions
- ♦ Catch bonds



Water is polar because of unequal  $e^-$  sharing:

$O = 3.5$  vs.  $H = 2.1$  ( $e^-$  negativity)

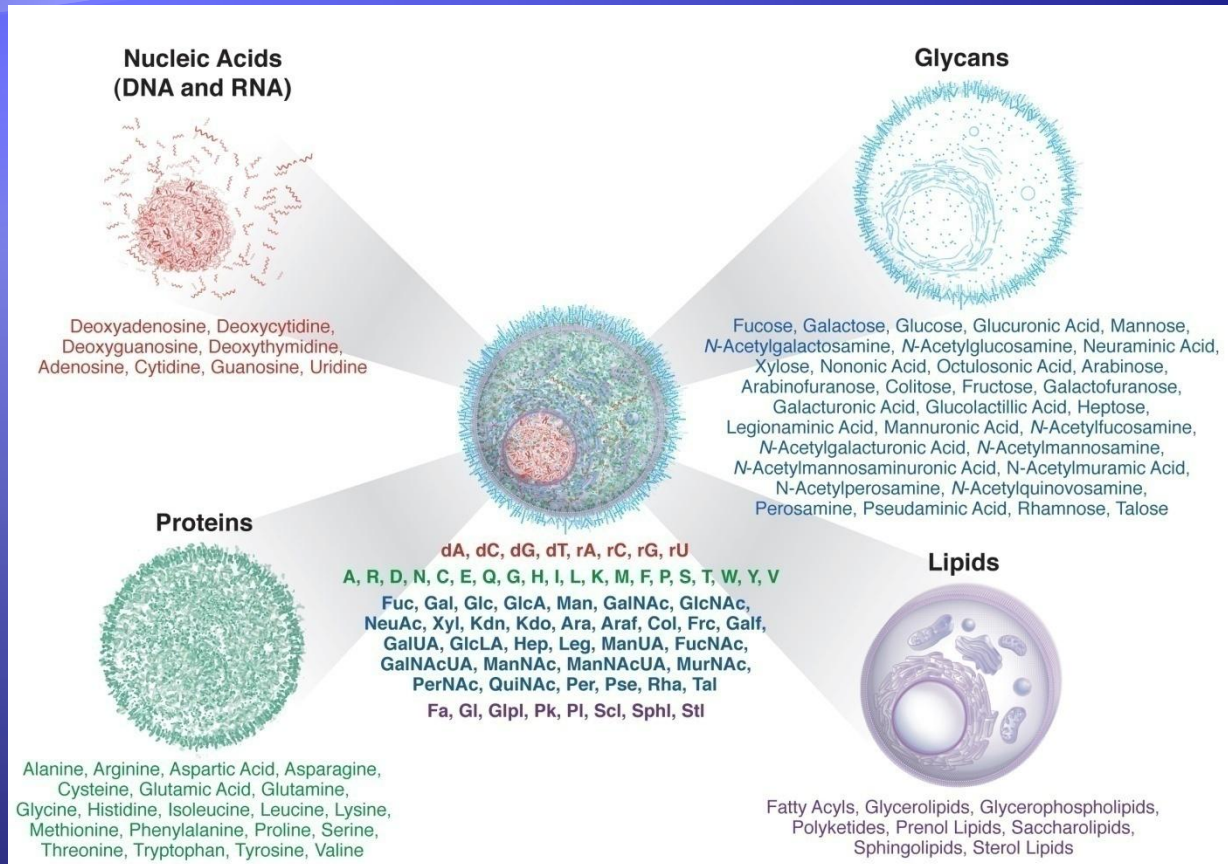
Solvents and hydrocarbon chains are nonpolar:

$C = 2.5$  vs.  $H = 2.1$

# Fundamental Units of Organisms

- ◆ All life is from 68 molecular building blocks

Nucleic Acids  
8 nucleosides



Glycans  
32+ sugars

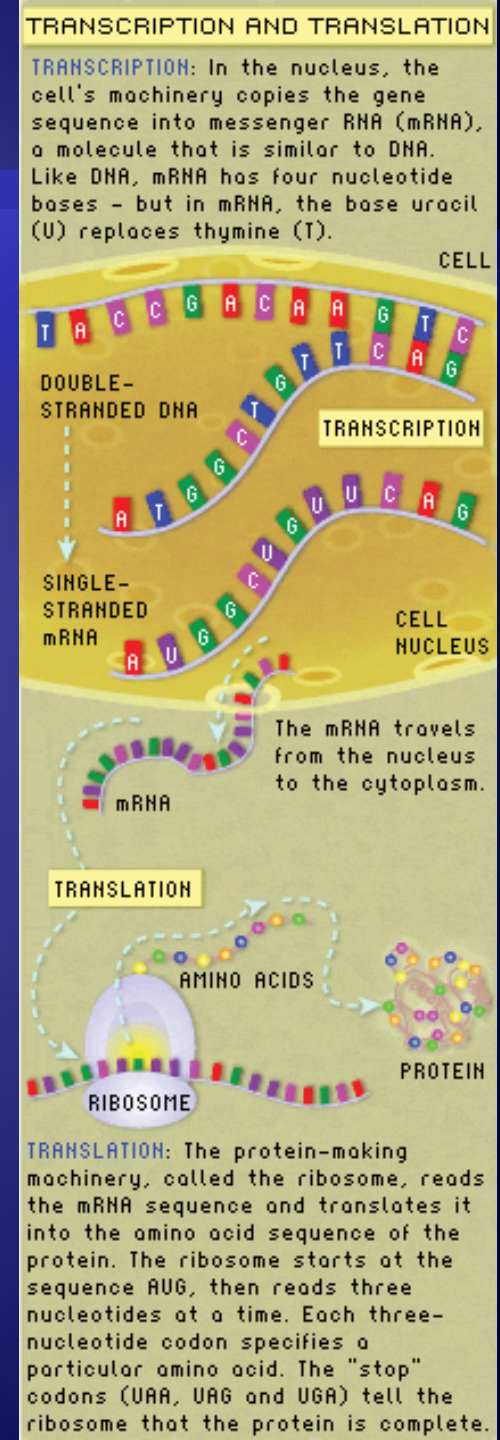
Proteins  
20 amino acids

Lipids  
8 types

“From the construction, modification, and interaction of these components, the cell develops and functions.” –James Marth

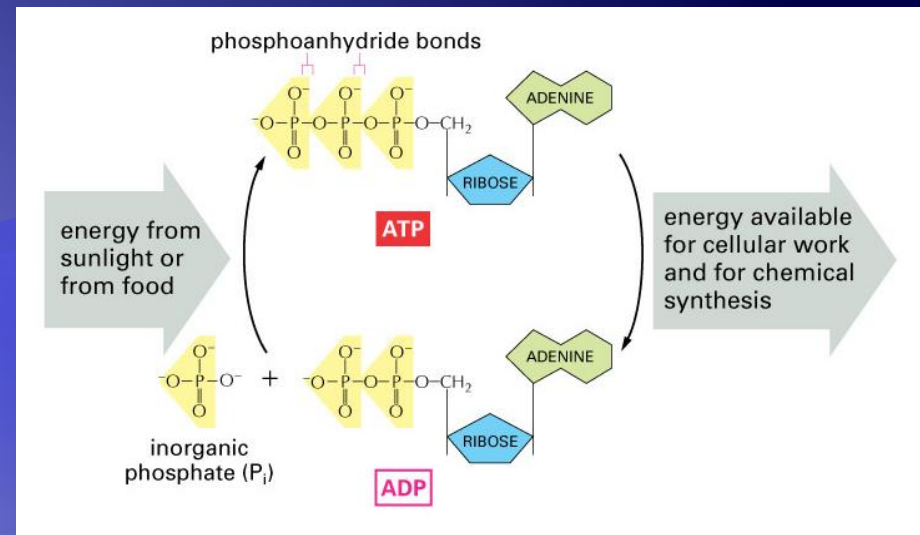
# Nucleic Acids

- ◆ DNA – genetic code
  - ◆ Adenine, Guanine, Cytosine, Thymine
- ◆ RNA – translation into proteins
  - ◆ Adenine, Guanine, Cytosine, Uracil
  - ◆ Messenger RNA (mRNA)
  - ◆ Transfer RNA (tRNA)
- ◆ Others...
  - ◆ Ribosomal RNA (rRNA)
  - ◆ Small interfering RNA (siRNA)
  - ◆ Micro RNA (microRNA)
  - ◆ Small nuclear RNA (snRNA)



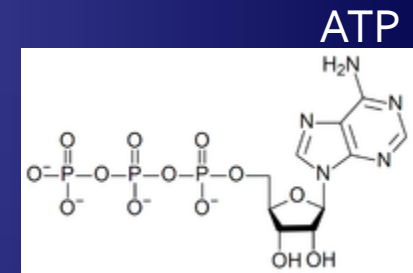


# Nucleoside Triphosphates



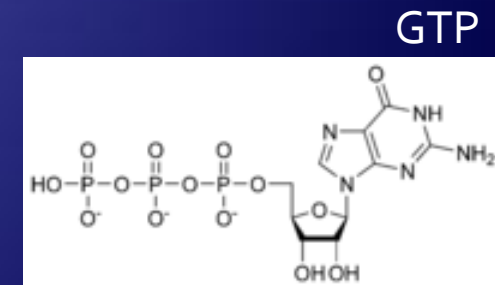
## ♦ ATP – energy currency

- ♦ Adenosine triphosphate
- ♦  $\text{ATP} + \text{H}_2\text{O} \rightarrow \text{ADP} + \text{PO}_4^{3-} + 7.3 \text{ kcal/mol}$
- ♦ Actin & Myosin

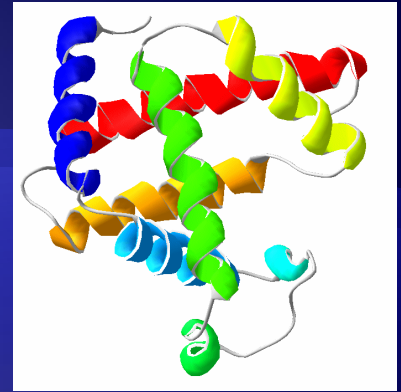


## ♦ GTP - regulatory

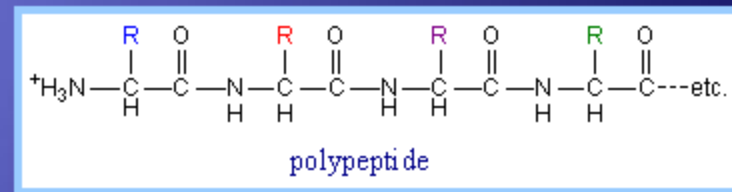
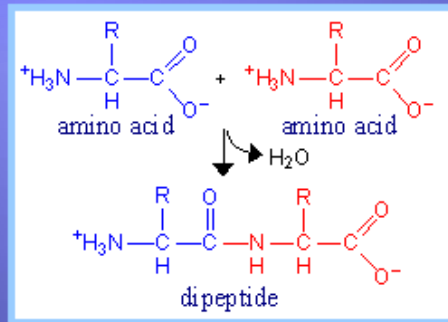
- ♦ Guanosine triphosphate
- ♦ G-protein Signal Transduction
- ♦ Microtubules



# Proteins



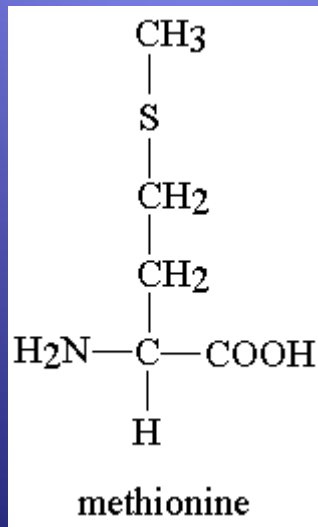
- ◆ Individual amino acids are translated into long chains called polypeptides
- ◆ Peptide bond: carboxyl + amino  $\rightarrow$  CO–NH + H<sub>2</sub>O



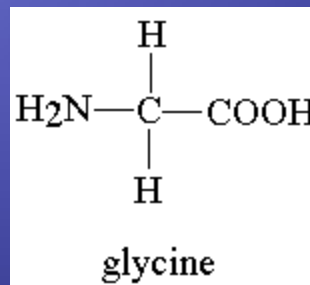
- ◆ Residue (R): each amino acid in a polypeptide
- ◆ Residue sequence read from N-terminal to C-terminal
- ◆ Final sequence will “fold up” into a 3-dimensional structure
- ◆ Substitution of just one residue can change a protein’s structure-function relationship

# Amino Acids

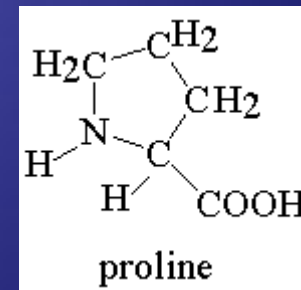
- ◆ Central carbon plus amino group ( $-\text{NH}_2$ ), carboxyl group ( $-\text{COOH}$ ), hydrogen atom ( $-\text{H}$ ), and distinct side chain (20 in total)
- ◆ Side chains determine shape and function



Met, M  
Start of all proteins



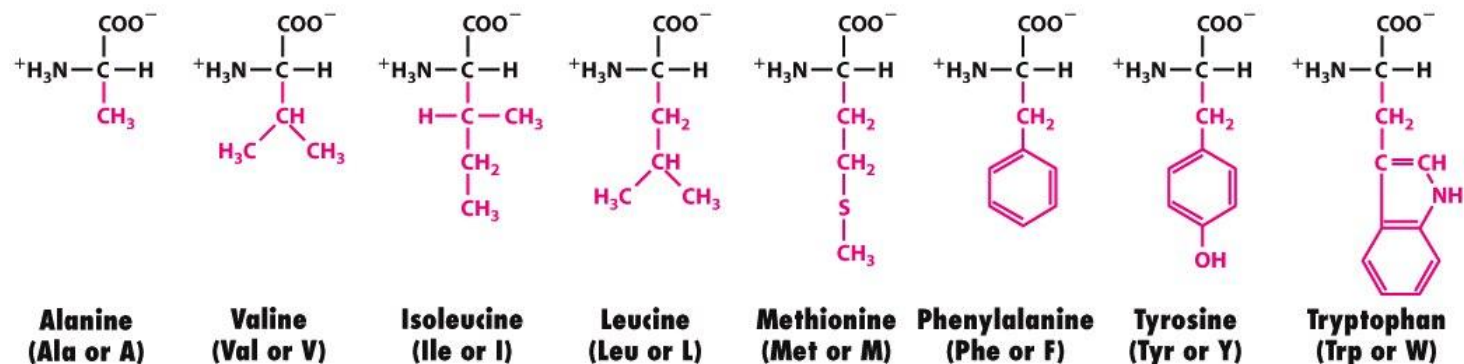
Gly, G  
Simplest  
Hydrophobic



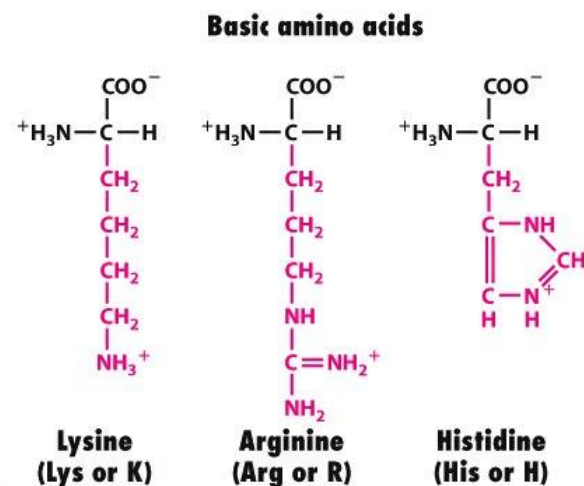
Pro, P  
Forms a rigid kink



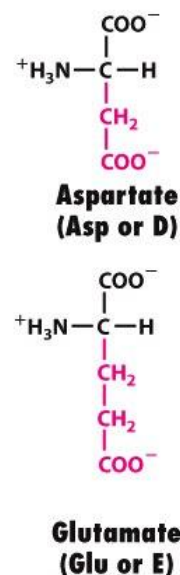
## HYDROPHOBIC AMINO ACIDS



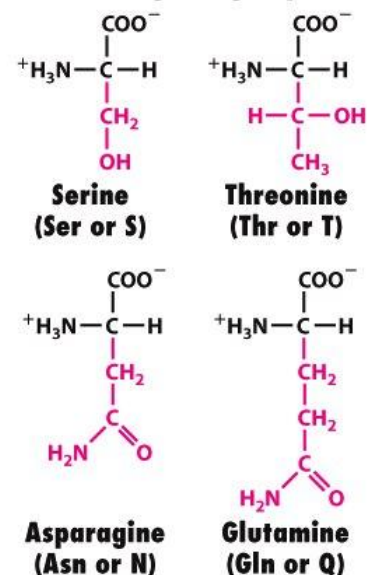
## HYDROPHILIC AMINO ACIDS



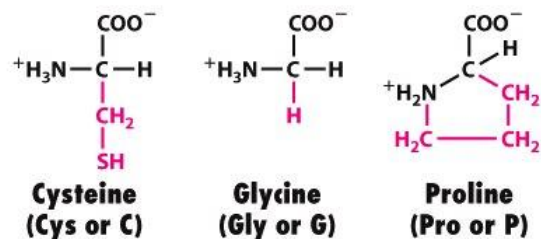
### Acidic amino acids



### Polar amino acids with uncharged R groups

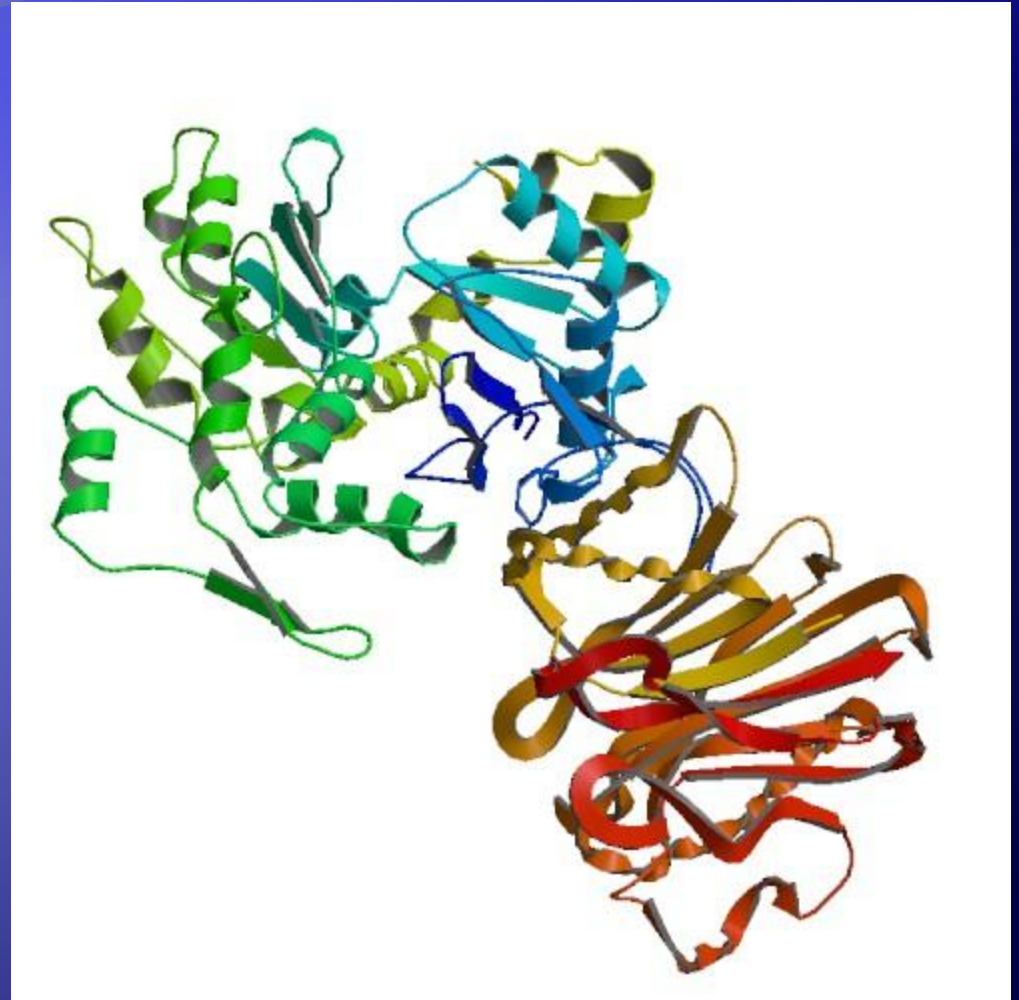


## SPECIAL AMINO ACIDS



# Amino Acid Sequence

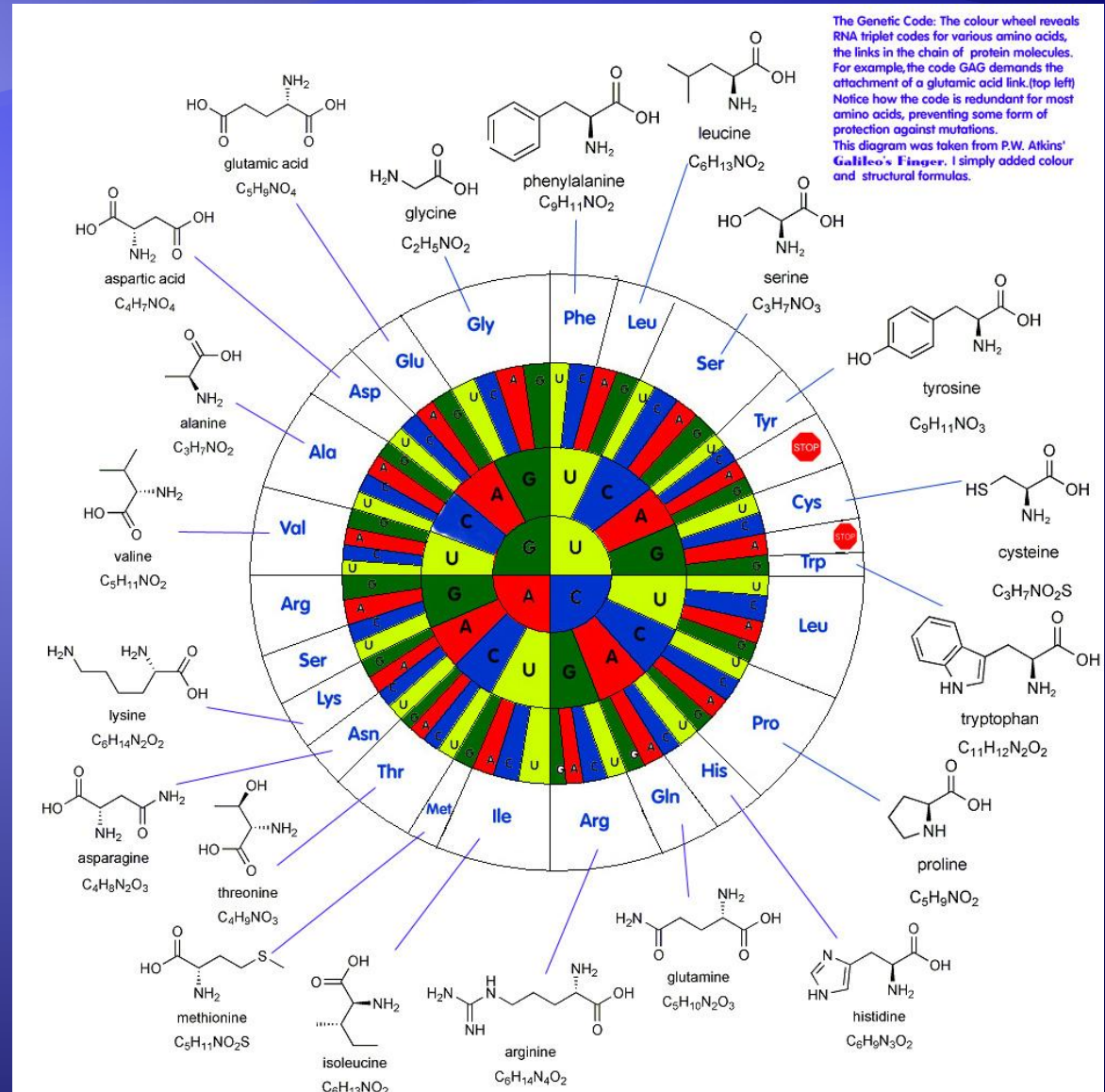
MCEEETTALVCDNGSGLCKAG  
FAGDDAPRAVFPSIVGRPRHQ  
GVMVGMGQDSYVGDEAQSK  
RGILTLYPIEHGIITNWDDME  
KIWHHSFYNELRVAPEEHPTLL  
TEAPINPKANREKMTQIMFET  
FNVPMYVAIQAVLSLYASGRT  
TGIVLDSGDGVTHNVPIYEGYA  
LPHAIMRLDLAGRDLTDYLMKI  
LTERGYSFVTTAEREIVRDIKEK  
LCYVALDFENEMATAASSSSL  
EKSYELPDGQVITIGNERFRCP  
ETLFQPSFIGMESAGIHETTYN  
SIMKCDIDIRKDLYANNVLSGG  
TTMYPGIADRMQKEITALAPS  
TMKIKIIAPPERKYSVWIGGSIL  
ASLSTFQQMWISKPEYDEAGP  
SIVHRKCF



SMOOTH MUSCLE ACTIN (ACTG<sub>2</sub>)

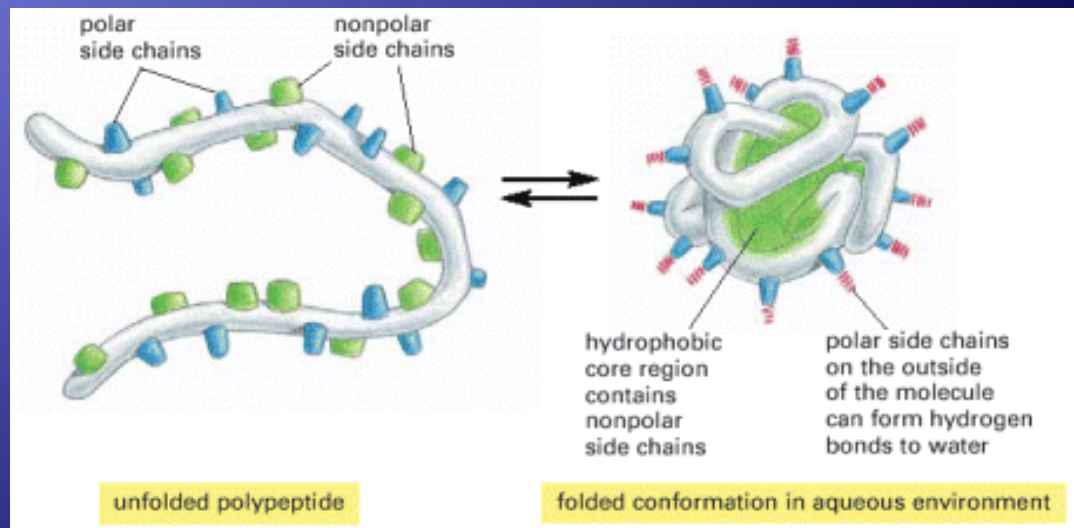
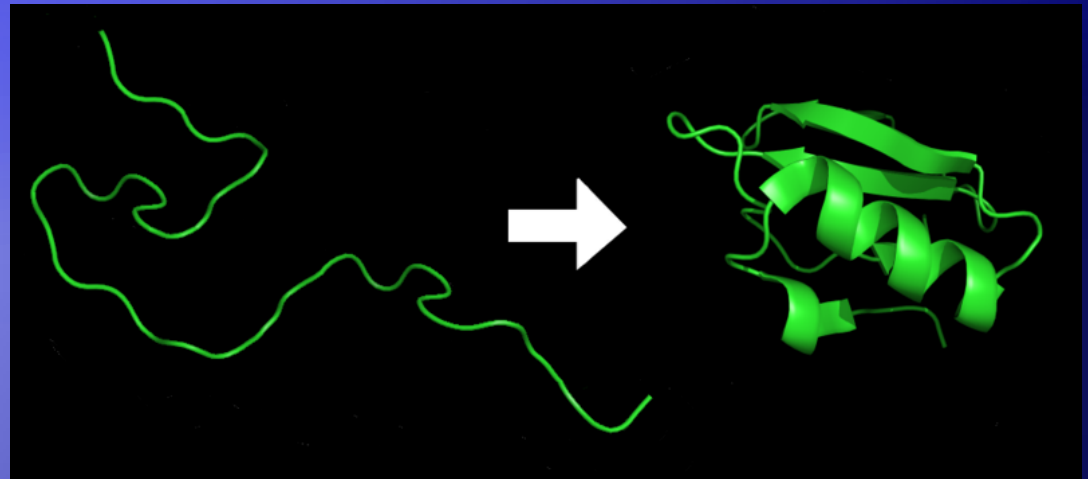
# Translation

- ◆ Start with Methionine (AUG)
- ◆ Build codon by codon
- ◆ End with UAA, UAG, UGA



# Protein Folding

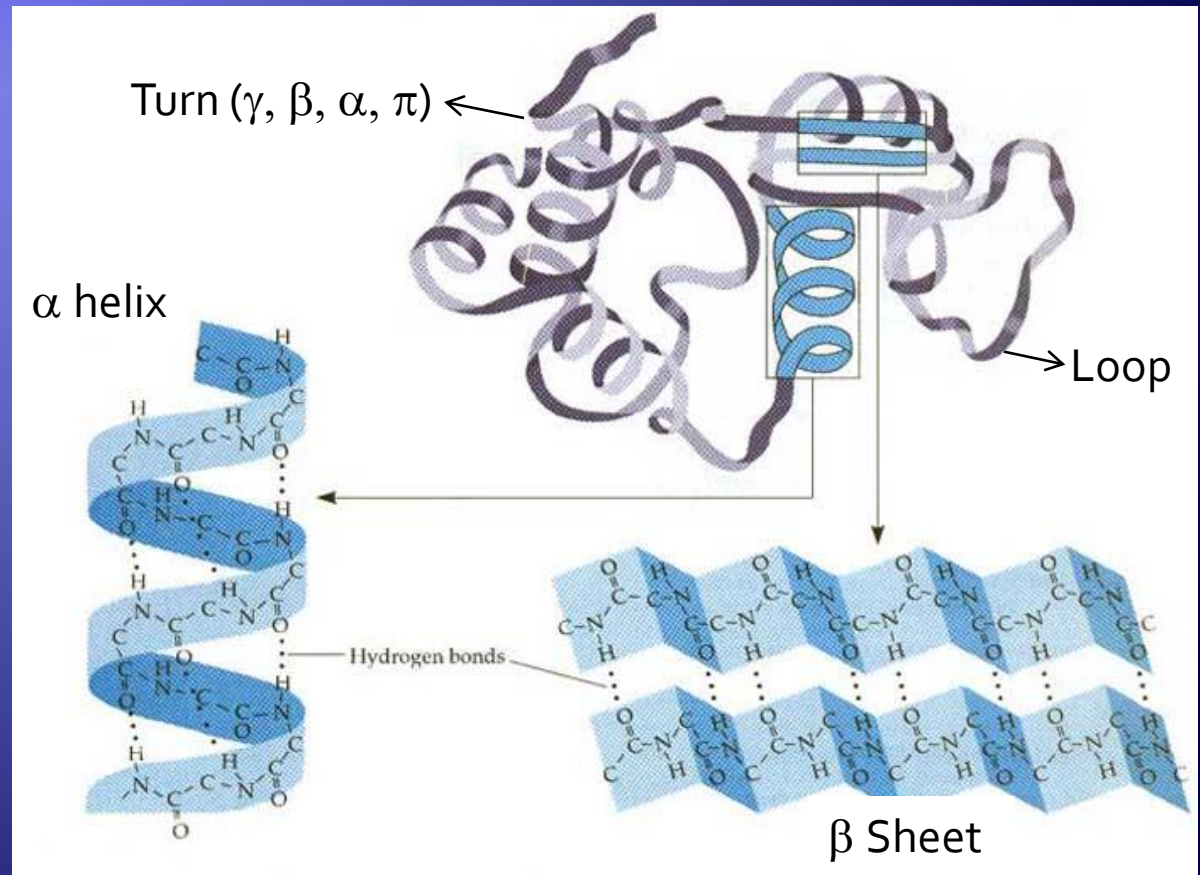
- ♦ Hydrophobic residues form inner core of protein
- ♦ Denaturing is loss of protein structure
- ♦ Post-translational modifications can change protein folding (...and change function)





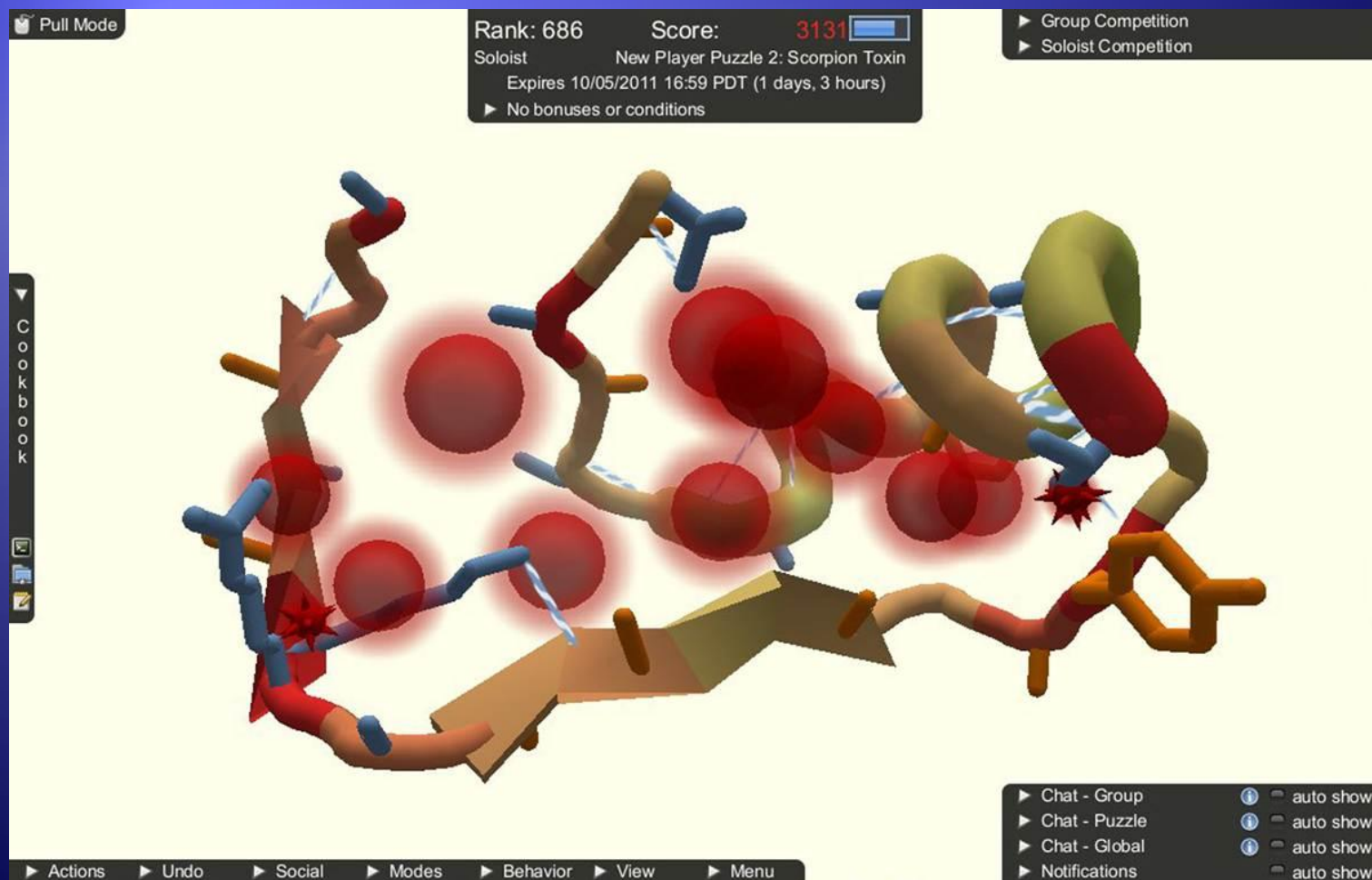
# Protein Structure

- ◆ Primary – amino acid sequence
- ◆ Secondary – sub-structures through hydrogen bonds
  - ◆ Alpha helix
  - ◆ Beta sheet
- ◆ Tertiary – overall shape of a single protein unit
- ◆ Quaternary – union of more than one protein units



# Foldit (Hw1)

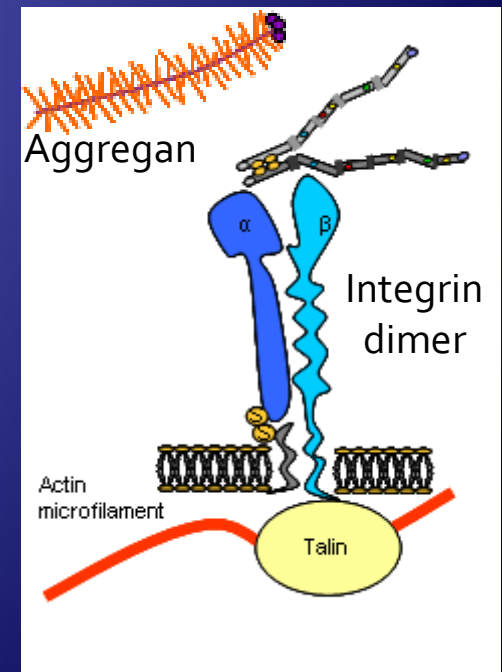
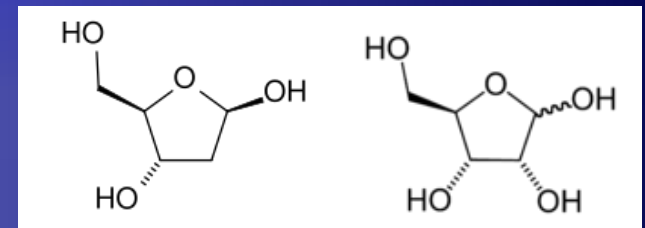
- ◆ Developed at UW by Seth Cooper from David Baker's Rosetta program





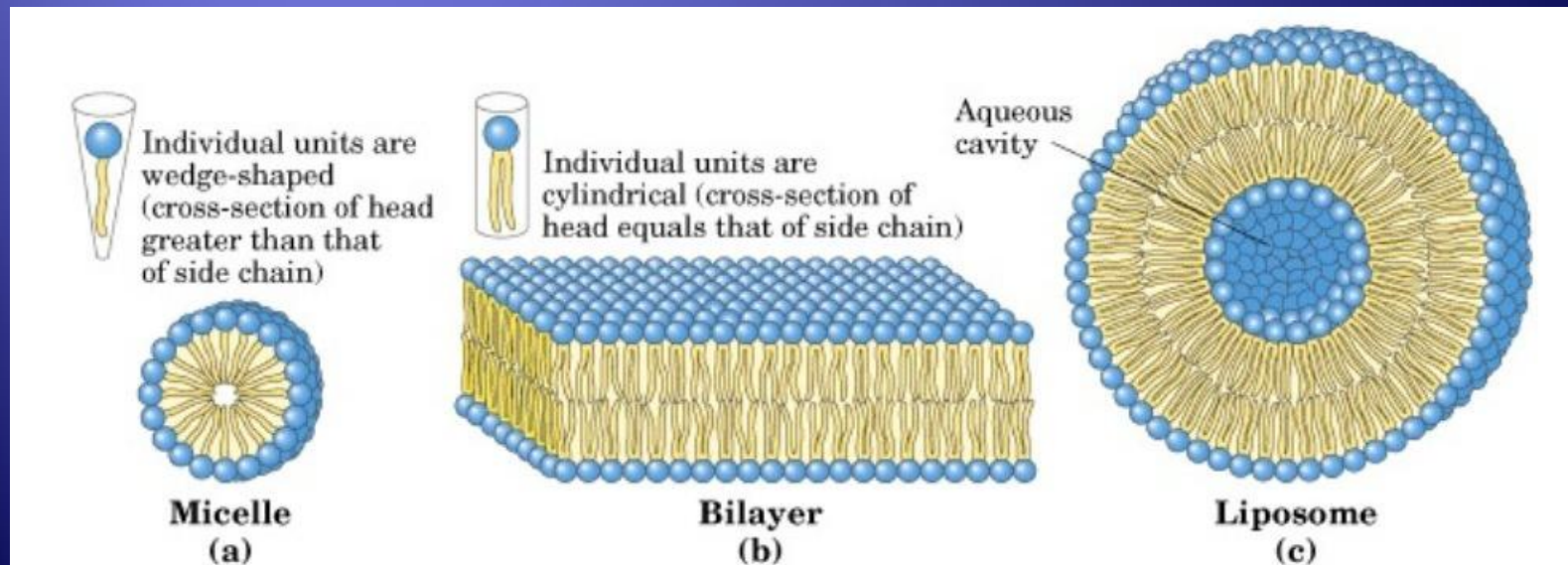
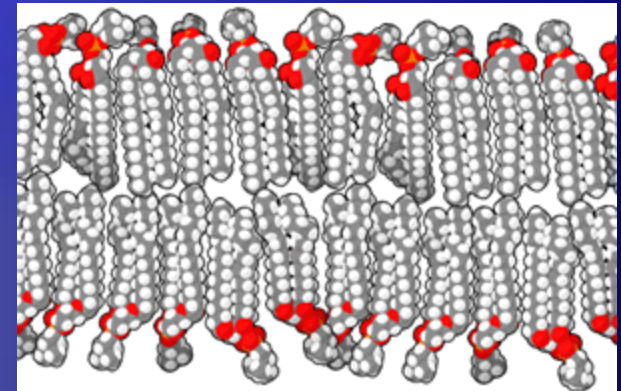
# Glycans

- ♦ “Carbohydrates” – hydrated carbons, (C, H, O)
- ♦ Combinatorial:
  - ♦ Glycoproteins – transmembrane proteins, e.g. integrins
  - ♦ Proteoglycans – protein core with polysaccharide chains, e.g. ECM’s aggrecan
  - ♦ Glycolipids – located in extracellular layer of the cell membrane

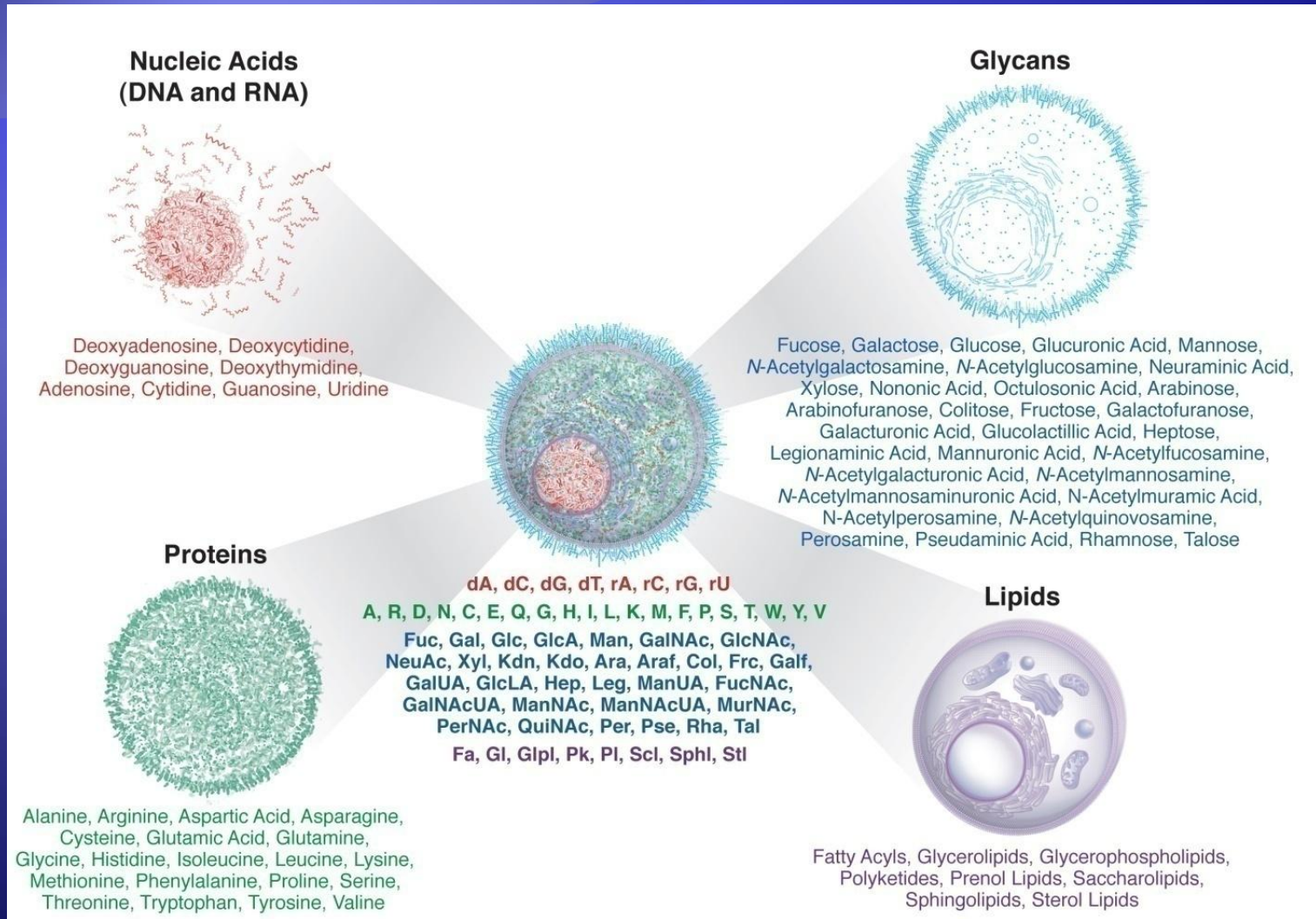


# Lipids

- ◆ Phospholipids have nonpolar hydrocarbon tails and negatively charged phosphate heads
- ◆ In water, phospholipids cluster together to form bilayers
- ◆ At air-water interface, tails point out and you have soap bubbles



# All Together Now...



Questions?