1	
2	Supporting Information for:
3	
4	Mass Balance Approaches to Characterizing the Leaching Potential of Trenbolone Acetate
5	Metabolites in Agro-Ecosystems
6	
7	Gerrad D. Jones <sup>1</sup> , Peter V. Benchetler <sup>1</sup> , Kenneth W. Tate <sup>2</sup> , Edward P. Kolodziej <sup>1*</sup>
8	
9	<sup>1</sup> Department of Civil and Environmental Engineering, University of Nevada-Reno, MS 0258,
10	Reno, Nevada, 89557
11	
12	<sup>2</sup> Department of Plant Sciences, University of California-Davis, MS 1, Davis, California, 95616-
13	8780
14	
15	For submission to Environmental Science and Technology
16	
17	*Corresponding author contact information:
18	Kolodziej, E.P; Telephone: (775) 682-5553; fax: (775) 784-1390; email: koloj@unr.edu
19	
20	
21	Contents: 19 pages, 5 Figures, 2 Tables.
22	
23	

## 24 Materials and Methods

25 *Chemicals*: 17α-TBOH (17α-hydroxyestra-4,9,11-trien-3-one), 17β-TBOH (17β-26 hydroxyestra-4,9,11-trien-3-one), and TBO (estra-4,9,11-trien-3,17-dione) were obtained from 27 Steraloids (Newport, RI). Deuterated 17β-TBOH (d3-17β-hydroxyestra-4,9,11-trien-3-one) was 28 obtained from the BDG Synthesis (New Zealand) and was used for isotope dilution recovery 29 correction for all TBA metabolites. HPLC-grade solvents were obtained from Fisher (Pittsburgh, 30 PA). Derivatization grade N-methyl N-trimethylsilyl-triflouroacetimide (MSTFA) and I<sub>2</sub> 31 (99.999% purity) were obtained from Sigma Aldrich (Milwaukee, WI). Complete descriptions for other experimental details can be found in Parker et al. (2012) and Webster et al (2012).<sup>1,2</sup> 32 33 Irrigation Leaching Mesocosms: Irrigation leaching mesocosms consisted of aluminum trays of three different areas (i.e., 120, 600, or 1200 cm<sup>2</sup>) filled with 1-2 L manure samples 34 35 evenly distributed across the bottom of each tray. Irrigation water was applied at a rate of 8 L/hr 36 through PVC pipes in series to each tray and the distance between the manure surface and the 37 spout was approximately 10 cm for each mesocosm. The total demand to the pipe system was 72 38 L/hr (9 trays x 8 L/hr), but the total flow through the pipe system was  $\sim 100$  L/hr. The excess 39 water flowed out of the system through an overflow pipe, which was the highest elevation of the 40 system at  $\sim 25$  cm above each tray. Because water continuously flowed out of the overflow pipe, 41 the water level at this point (i.e., ~25 cm) effectively fixed the pressure head throughout the 42 entire system, and because we controlled the flow to each tray through individual valves, the 43 flow to each tray was constant throughout the duration of the experiment, despite fluctuations in 44 pressure from the feed irrigation water (Figure S1). 45 Sample processing and analysis: All samples were extracted onto solid phase extraction

46 (SPE) cartridges. The flow through each sample was < 10 mL/minute for both loading and

47 elution steps. If necessary, SPE cartridges were stored in a 1°C refrigerator prior to elution and 48 analysis. SPE cartridges were eluted with methanol:water (9 mL, 95:5 v/v). The eluent was 49 dried, resuspended in dichloromethane:methanol (12 mL, 95:5 v/v), purified using Florasil 50 cartridges (6 mL, Restek), and dried to ~1 mL. The eluent was transferred to 2 mL vials, dried 51 under nitrogen, derivatized using MSTFA-I<sub>2</sub> (50  $\mu$ L; 1.4 mg I<sub>2</sub>/mL MSTFA), and immediately 52 dried again to remove residual iodine. Finally, extracts were resuspended in MSTFA (100 µL), 53 heated at 60°C (40 min), and cooled to room temperature prior to analysis. Samples were 54 analyzed by GC/MS/MS (Agilent 6890N, Santa Clara, CA, USA; Waters Quattro Micromass 55 spectrometer, Milford, MA, USA).

56 In addition to TBA metabolites, irrigation leaching samples were analyzed for ammonia, 57 nitrite, nitrate, orthophosphate, total coliforms, E. coli, and total organic carbon using standard 58 approaches. Following filtration with a  $0.7 \,\mu m$  glass fiber filter sample (see text), aliquots of the 59 4 L leachate samples were filtered through an additional 0.45 µm glass fiber vacuum filter. TOC 60 samples were acidified to pH = 2 using phosphoric acid and analyzed using a Shimadzu TOC 61 combustion analyzer. Ammonia, ammonium, nitrite, nitrate, and orthophosphate were analyzed 62 using flow injection analysis on a Lachat Quickchem 8500 auto analyzer. Total Coliform and E. 63 *coli* samples were collected in sterilized 120 mL plastic vessels purchased from IDEXX 64 Laboratories. MPN concentrations were determined using the 'Colilert' defined substrate 65 method from IDEXX. The IDEXX 51 well travs were filled with 100 mL of diluted sample (1:100,000 dilution) mixed with the 'Colilert' synthetic media, sealed using the Quanti-tray<sup>TM</sup> 66 67 sealer, and incubated at 35 °C for 24 hours. Positive detect wells were counted and compared with standard Quanti-tray<sup>TM</sup> MPN tables. 68

69

## 71 Results and Discussion

*Diffusion Model Parameter Estimation*: Using mechanistic approaches first derived for
 sediment systems, we assumed manure could be modeled as saturated porous media and used the
 following one-dimensional diffusion model to describe the mass flux of TBA metabolites from
 manure:

76 
$$L(t) = \left(\frac{4D}{\pi f}\right)^{1/2} \varphi C_w t^{1/2} \qquad (1)$$

where L(t) is the area normalized mass leached (ng/cm<sup>2</sup>), D is the effective diffusivity of TBA metabolites in manure (cm<sup>2</sup>/s), f is the dissolved fraction of TBA metabolites,  $\varphi$  is the porosity (V<sub>voids</sub>/V<sub>total</sub>), C<sub>w</sub> is the aqueous equilibrium concentration of steroids in manure (ng/cm<sup>3</sup>), and t is time (s).<sup>3</sup> A description of how each variable was estimated is described below.

81 D (cm<sup>2</sup>/s): The effective diffusivity of  $17\alpha$ -TBOH in fresh-manure was estimated from the free 82 diffusivity in water (D<sub>w</sub>). We used three empirical formulas to estimate D<sub>w</sub> in water at a 83 minimum (T = 0°C) and maximum (T = 37°C) temperature. These formulas are as 84 follows:

85 
$$D_w = \frac{14.0(10^{-5})}{\mu^{1.1}V^{0.6}}$$
 (2; Othmer and Thakar, 1953)<sup>4</sup>

86 
$$D_w = \frac{7.4(10^{-8})(2.6 M)^{0.5}T}{\mu V^{1/3}}$$
 (3; Wilke and Chang, 1955)<sup>5</sup>

87 
$$D_w = \frac{13.26(10^{-5})}{\mu^{1.4}V^{0.589}} \qquad (4; \text{ Hayduk and Laudie, } 1974)^6$$

where µ is the viscosity (centipoise), V is the molar volume (cm<sup>3</sup>/g), and T is temperature
(°K). We assumed that fresh manure was fully saturated with water and used the
following equation to estimate the effective diffusivity:

91 
$$D = D_w \varphi^{\frac{4}{3}}$$
 (5; Millington and Quirk, 1961)<sup>7</sup>

92 φ (%): We estimated φ by oven drying 5 ml of fresh manure at 105° C for 24 hours and dividing
93 the difference between the wet and dry manure (i.e., the volume of water assuming 1 g =
94 1 cm<sup>3</sup>) by the total volume (i.e., 5 ml). The water content of fresh manure ranged from
95 0.81-0.86 and was 0.83 ± 0.01 (± 95% confidence interval) on average (n = 27).

96 f(%): The dissolved fraction of TBA metabolites in manure was calculated as follows:

97 
$$f = \frac{1}{1 + rK_D} \tag{6}$$

98 where  $K_D$  is the solids to water partitioning coefficient (cm<sup>3</sup>/g), which was estimated 99 using the 17 $\alpha$ -TBOH partitioning coefficient to organic carbon (i.e.,  $K_{oc}$ )<sup>8</sup> and the 100 fraction of organic carbon ( $f_{oc}$ ). We estimated  $f_{oc}$  based on the fraction of organic matter 101 ( $f_{om}$ ) in manure using ASTM method D2974-07a (loss on ignition) and using the Van 102 Bemmelen factor (i.e.,  $f_{oc} = 58\% f_{om}$ ). The fraction of organic matter in dried fresh-103 manure ranged from 0.77-0.85 and was  $0.81 \pm 0.01$  on average (n = 27).

104 r is the ratio of solids to water  $(g/cm^3)$  and was calculated as follows:

105 
$$r = \rho \frac{1-\varphi}{\varphi} \qquad (7)$$

106 where  $\rho$  is the dry manure density (g/cm<sup>3</sup>), which we estimated by oven drying 5 ml of 107 wet manure at 105° C for 24 hours and dividing the dried mass by the total volume (i.e., 5 108 ml). The density of dry manure ranged from 0.14-0.21 g/cm<sup>3</sup> and was 0.18 ± .01 g/cm<sup>3</sup> 109 on average (n = 27) and the density of wet manure ranged from 0.90-1.17 g/cm<sup>3</sup> and was 110  $1.03 \pm 0.01$  g/cm<sup>3</sup> on average (n = 28).

never detected TBA metabolites within the provided irrigation water, the concentration
gradient was simply the equilibrium concentration of TBA metabolites within the
manure. We estimated the aqueous concentration in manure using the following mass
balance:

$$S_t = C_w V_w + C_s M_s \quad (8)$$

where  $S_t$  is the total metabolite mass excreted in a sample (ng),  $C_w$  is the aqueous equilibrium concentration of TBA metabolites within the sample (ng/cm<sup>3</sup>),  $V_w$  is the sample volume of water in manure (cm<sup>3</sup>),  $C_s$  is the solid concentration of metabolites (ng/g-dw), and  $M_s$  is the mass of solids in manure (g-dw). For a 1 g-dw sample, the previous mass balance can be written as:

$$C_m = C_w V_w + C_s \times 1 \tag{9}$$

125 where  $C_m$  is the TBA metabolite concentration in manure (ng/g-dw). It should be noted 126 that  $C_m$  and  $C_s$  have the same units (i.e., ng/g-dw) but do not represent the same 127 parameter.  $C_m$  describes the total mass of TBA metabolites, both dissolved and sorbed, 128 in fresh manure that has been dried and  $C_s$  describes the mass of TBA metabolites that 129 are sorbed to solids resulting from equilibrium partitioning.  $K_D$  and  $C_w$  were substituted 130 into  $C_s$  (i.e.,  $K_d*C_w = C_s$ ) and the equation was rearranged and solved for  $C_w$  as follows:

131 
$$C_w = C_m \left(\frac{1}{V_w + K_D \times 1}\right) \quad (10)$$

132This form is advantageous for substitution into the diffusion model because the  $C_m$  is133often reported in the literature while  $C_w$  is not. The volume of water in 1 g-dw equivalent134of fresh manure (~5.88 g-ww) can be described as follows:

135 
$$V_w = 1 g(dw) \times \frac{1 g(ww)}{(1-\varphi) g(dw)} \times \frac{\varphi g H_2 0}{1 g(ww)} \times \frac{1 cm^3 H_2 0}{1 g H_2 0}$$
(11)

Each grouping represents a unit conversion. For example, in the second group, 1 g-ww
of manure is equivalent to 1-φ g-dw of manure (~0.17 g-dw), and in the third grouping, 1

138 g-ww of manure is equivalent to  $\varphi$  g of water (~0.83 g water), etc. Therefore, the 139 aqueous concentration of TBA metabolites in manure can be written as follows:

140 
$$C_w = C_m \left(\frac{1}{\frac{\varphi}{(1-\varphi)} + K_D \times 1}\right)$$
(12)

We used a Monte Carlo simulation (n = 10,000, see below) to estimate C<sub>w</sub> based C<sub>m</sub>
 ranging from 0-70 ng/g-dw (4-64 ng/g-dw was observed in experiments) by selecting
 random values from within the measured, calculated, or reported range of each

144 independent variable (Table S1). The average product of 
$$\left(\frac{1}{\frac{\varphi}{(1-\varphi)}+K_D\times 1}\right)$$
 was 0.0034 ±

145 0.0007 g-dw/cm (± stdev; Figure S4). Therefore, we simplified the mass balance to
146 describe C<sub>w</sub> as follows:

147 
$$C_w = 0.0034C_m$$
 (13)

148 This mass balance was used to simplify the 1-D diffusion model.

149

*Monte Carlo Statistical Analysis:* Within the diffusion model,  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  is a constant for 150 151 a given contaminant within a given media. We used the Monte Carlo simulation described within the text to determine whether the observed value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  obtained from the 152 153 simulated leaching experiment was statistically different from that of the average of the model. We estimated  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  from the each simulated-irrigation leaching experiment by dividing the 154 area normalized mass leached (L) at each time point by  $C_w$  and  $t^{1/2}$ . For 17 $\alpha$ -TBOH sample 155 concentrations of 24 and 63 ng/g-dw, the average value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  was 0.0055 and 0.0070 156 cm/s<sup>0.5</sup>, respectively. Based on the Monte Carlo simulations, the average value was 0.0065 157

158 cm/s<sup>0.5</sup>. We created a cumulative probability distribution describing the probability that a 159 randomly generated value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$ , from each iteration, was less than a particular value of x, 160 namely 0.0055 and 0.0070 cm/s<sup>0.5</sup>. The cumulative probability (P<sub>c</sub>) that a random value is  $\leq x$ 161 ranges from 0 < P  $\leq$  1 and can be described as follows:

162 
$$P_c = \frac{\# of \ values \le x}{10,000}$$
 (14)

174

For example, if x is the maximum value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  for all 10,000 iterations (i.e., 0.01 163 cm/s<sup>0.5</sup>), the cumulative probability that a randomly chosen value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  is  $\leq x$  is 1.0000 164 (i.e., P = 10,000/10,000). Conversely, if x is the minimum value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  for all 10,000 165 iterations (i.e.,  $0.003 \text{ cm/s}^{0.5}$ ), the cumulative probability that a randomly chosen value of 166  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  is  $\leq$  x is 0.0001 (i.e., P = 1/10,000). Therefore, the cumulative probability that a 167 randomly chosen value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  is less than the average value (i.e., 0.0065 cm/s<sup>0.5</sup>) is 0.5 168 169 since 50% of the observations are below and 50% of the observations are above this value. Therefore, the probability that any value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  is statistically different from the average 170 171 can be described as follows: 172  $P = 2(0.5 - |0.5 - P_c|)$ (15) 173 For example, for a particular value x with  $P_c = 0.5$  (i.e., x = the mean value), P = 1.0000,

175 mean value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$ ). For P<sub>c</sub> < 0.025 or P<sub>c</sub> > 0.975 (i.e., the upper and lower 2.5% of all 176 10,000 observations), P < 0.05, and we would reject the null hypothesis in favor if the alternative

and we would fail to reject the null hypothesis (i.e., there is no difference between x and the

hypothesis (i.e., there is a difference between x and the mean value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$ ). For 0.0055 177 and 0.0070 cm/s<sup>0.5</sup>, P<sub>c</sub> = 0.2884 and 0.6461, respectively. Therefore, the probability that there is 178 no difference between the estimated value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  from both leaching experiments and the 179 mean value from all 10,000 iterations (i.e., 0.0065 cm/s<sup>0.5</sup>) is P = 0.5768 (for 24 ng/g-dw) and P 180 = 0.7078 (for 63 ng/g-dw). Because there was no statistical difference between the observed and 181 182 modeled leaching (i.e., P > 0.05), and because the observed and modeled leaching were 183 independent (i.e., no leaching data was used to build the model), the diffusion model was 184 deemed appropriate to describe 17a-TBOH leaching.

185 Rainfall Leaching Model: Rainfall leaching was highly correlated with rainfall intensity, 186 TBA metabolite concentration, and rainfall depth (see main text). Because rainfall leaching is 187 inherently correlated with both the sample concentration as well as some rainfall metric, each 188 model contains two independent variables describing a particular storm event and the 189 concentration. More variables were not used to prevent over-parameterization of the model. 190 Given the absence of available mechanistic approaches, we evaluated three empirical regression 191 models to describe the mass of TBA metabolites that could leach (ng/kg-dw) during rainfall 192 events (L<sub>r</sub>):

193 
$$\log(L_r(I, C_m)) = 1.16I + 0.022C_m - 1.99$$
 (16); R<sup>2</sup> = 0.97

194

$$L_r = 54.26 (IC_m)^{1.115}$$
(17); R<sup>2</sup> = 0.88

195  $L_r = 11.34 (DC_m)^{1.147}$  (18);  $R^2 = 0.79$ 

where I is the maximum hourly rainfall intensity (cm/hr), C<sub>m</sub> is the TBA metabolite
concentration in manure (ng/g-dw), and D is the rainfall depth (cm). While each model is
empirical, we evaluated the models based on two criteria: does the model make physical sense,
especially at the environmental limits of each independent parameter, and which model is the

200 most parsimonious. Because the mass leached during rainfall events varied across three orders 201 of magnitude (i.e., 5-2500 ng), we log transformed the leaching data to prevent extreme events 202 from driving the multiple linear regression model (equation 16). While equation 16 accounted 203 for the greatest variability within the dataset, the mass leached is always greater than 0 ng, 204 regardless of the value the rainfall intensity and concentration. It is also particularly sensitive to 205 high rainfall intensities, especially when concentrations are low. For example, at 0.1 ng/g-dw, 206 but the intensity is 2.5 cm/hr, the mass leached is 8 ng. For a 5 cm/hr intensity, the mass leached 207 is 6,500 ng/g-dw. The model is less sensitive when concentration is high but intensity is low. 208 Equation 16 over estimates the mass that can leach when both concentration and intensity are 209 low, which is of particular concern since the average concentration of TBA metabolites falls 210 relatively quickly following implantation (see text). While equation 16 fit the data well, it did 211 not make physical sense for low values of intensity and concentration and was therefore removed 212 for model consideration.

213 Power-law relationships have been used to describe contaminant leaching during rainfall events.<sup>7</sup> Therefore, we developed two power models to describe leaching based on the 214 215 concentration and intensity or the concentration and rainfall depth. In order to ensure that the 216 mass leached approached zero ng as either independent variable approached zero, we generated a 217 single composite variable that was the product of the maximum hourly rainfall intensity and the 218 sample concentration. While both models (equations 17 and 18) approach 0 ng leached as either 219 independent variable approaches zero, the rainfall intensity in equation 17 is more difficult to 220 estimate accurately than rainfall depth. While the intensity can be estimated from depth using 221 SCS storm curves based on the storm type (e.g., I, Ia, II, and III) and storm duration (e.g., 6, 18, or 24 hour storm), we considered equation 18 to be more parsimonious despite the lower  $R^2$ 222

- value because it measures the leachable mass directly. Equation 18 was therefore deemed more
- suitable than the equations 16 and 17 because: 1) the mass leached approaches zero as either
- rainfall depth or concentration approach zero, and 2) rainfall depth is easier to estimate *a priori*
- 226 for a particular event than rainfall intensity.
- 227



229 Figure S1. Irrigation (a and b) and rainfall (c) leaching mesocosms. Irrigation water was applied 230 to each tray  $(1,200 \text{ cm} 2 \text{ [a]} \text{ and } 120 \text{ and } 600 \text{ cm}^2 \text{ [b]})$  at a rate of 8 L/hr (as seen in [a] and [b]). 231 The leachate from each mesocosm overflowed directly into 4 L amber glass bottles. The 232 pressure head of the system was fixed by an overflow pipe (b) which was the highest elevation of 233 the piping system. The direction of flow was from the bottom right to the top left (b). Rainfall 234 mesocosms consisted of 12 L stainless steel pots (c). Samples were placed within aluminum 235 screen cylinders (23 cm diameter, 5 cm height, seen in all 4 pots) that were open to the top to 236 allow rainfall to impact the manure surface but prevent rainfall induced erosion of the sample. 237 The screens also maintained a near constant top interfacial area (410 cm<sup>2</sup>/kg-ww) exposed to 238 rainfall. Screens and samples were suspended on a stainless steel plates (seen in lower left of 239 [c]). After each rainfall event, leachate was collected from the bottom of each mesocosm. 240



243 **Figure S2.** Observed leaching of (a)  $17\alpha$ -TBOH, N, P, and TOC, and (b) total coliforms and E. 244 *coli* from manure during simulated irrigation leaching experiments. Nitrite and nitrate were 245 below the limits of detection and not presented. All data were collected from the same experiment, and the  $17\alpha$ -TBOH data corresponds to the data presented in Figure 3a (see text). 246 247 Note the differences in units within and between the figures. The primary y-axis represents the 248 normalized (both area and concentration) leaching and the secondary y-axis represents that mass 249 that can leach from 40 kg-ww of manure, which is the mass excreted from a single adult animal 250 unit (AU) over 24 hours, assuming complete manure submersion (i.e., 860 cm<sup>2</sup>/kg-ww x 40 kgww/AU = 34,400 cm<sup>2</sup> interfacial area). In (b), concentrations were multiplied by the entire 251 252 sample volume (4,000 mL) as well as the total submerged interfacial area to obtain an estimate of 253 the coliform forming units (CFU) that can leach from samples. Initial concentrations for 254 contaminants except 17 $\alpha$ -TBOH (24 ng/g-dw) were not extracted from the manure samples and 255 measured directly, thereby precluding the use of the diffusion model described in equation 1. 256 Error bars represent 95% confidence intervals. 257



Figure S3. Cumulative probability distribution of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  generated from the Monte Carlo analysis (n = 10,000 iterations). The outer 5% of all iterations (i.e., the greatest 250 (P<sub>c</sub> > 0.95) and smallest 250 (P<sub>c</sub> < 0.025) values of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$ ) are indicated by the dotted lines and were used to define a rejection region. If  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  fell within this rejection region, the null hypothesis (i.e., there is no difference between the observed value and the average) was rejected at P < 0.05. The probability (P) that the observed value of  $\left(\frac{4D}{\pi f}\right)^{1/2} \varphi$  is statistically different from the average for is indicated for both simulated irrigation leaching experiments.





**Figure S4.** Relationship between the total  $17\alpha$ -TBOH concentration in manure (C<sub>m</sub>; ng/g-dw) and the corresponding aqueous equilibrium concentration (C<sub>w</sub>; ng/mL). Using a Monte Carlo simulation (n = 10,000 iterations), values of C<sub>m</sub> ranging from 0-70 ng/g-dw (we observed 4-64 ng/g-dw in actual samples) were generated. For each C<sub>m</sub> value, the corresponding C<sub>w</sub> value was estimated from the total mass balance described previously (equations 8-12). The slope of the

273 line is the average product of 
$$\left(\frac{1}{\frac{\varphi}{(1-\varphi)}+K_D\times 1}\right)$$
 in equation 12



275

276 Figure S5. From animals implanted with 40 mg TBA, closed lines (primary y-axis) represent 277 the average concentration of  $17\alpha$ -TBOH in manure excreted onto the land surface through t days 278 post implantation. The average manure concentration is a function of the first-order 279 transformation rate constant, which was 0.17, 0.26, and 0.44/d at 1, 19, and 33°C, respectively. 280 These values represent environmentally realistic average seasonal temperature ranges for 281 rangelands and pastures. The mass of 17α-TBOH that can accumulate on the land surface (closed 282 lines; secondary y-axis) is a function of both the  $17\alpha$ -TBOH concentration (ng/g-dw) and the 283 manure production per animal unit (AU; g-dw/AU; equation 2 of main text). The mass of  $17\alpha$ -TBOH on the land surface peaks around 30 days post implantation. 284 285

Parameter Min Max Source  $^{\dagger}D_{w}$  (cm<sup>2</sup>/s) \*Hayduk and Laudie (1974)<sup>4</sup> 2.4E-06 9.3E-06  $D(cm^2/s)$ 1.8E-06 7.6E-06 Derived <sup>†</sup> $\rho$  (kg-dw/m<sup>3</sup>) 0.14 0.21 Measured  $^{\dagger}\phi$  (unitless) 0.81 0.86 Measured  $r (g/cm^3)$ .032 .035 Derived f (unitless) 0.13 0.07 Derived  $^{\dagger}K_{oc}$  (cm<sup>3</sup>/g) Khan et al.  $(2009)^8$ 832 447 <sup>†</sup>f<sub>om</sub> (unitless) 0.77 0.85 Measured  $f_{oc}/f_{om}$  (unitless) 0.58 0.58 Van Bemmelen factor foc (unitless) 0.49 Derived 0.45  $K_d$  (cm<sup>3</sup>/g) 199 410 Derived

\*values from Othmer and Thakar (1953)<sup>2</sup> and Wilke and Chang (1955)<sup>3</sup> fell

 Table S1. Range of measured, derived, or estimated values for diffusion

model independent variables.

within the range of Hayduk and Laudie (1974)<sup>4</sup> <sup>†</sup>variables randomized in Monte Carlo analyses.

**Table S2.** Five NOAA rain gages were used to estimate the rainfall

 depth and intensity at SFREC. The gage location, distance to SFREC,

 inverse squared distance, and the weight (derived from inverse distance

 weighting) are included.

	Distance to				
City	northing	easting	SFREC (km)	$1/D^2$	Weight
Auburn	4313330	666199	37.7	7.0E-04	0.09
Oroville	4371916	618857	37.9	7.0E-04	0.09
Beale	4333028	635136	15.3	4.3E-03	0.58
Yuba City	4328589	623622	26.8	1.4E-03	0.19
Emigrant Gap	4349751	697562	52.7	3.6E-04	0.05
SFREC	4344609	645113			
<u> </u>			Σ	7.4E-03	1.00

## 289 Supplemental References

Parker, J. A.; Webster, J. P.; Kover, S. C.; Kolodziej, E. P., Analysis of trenbolone
 acetate metabolites and melengestrol in environmental matrices using gas chromatography–
 tandem mass spectrometry. *Talanta* 2012, *99*, 238-246.

- 293 2. Webster, J. P.; Kover, S. C.; Bryson, R. J.; Harter, T.; Mansell, D. S.; Sedlak, D. L.;
- 294 Kolodziej, E. P., Occurrence of Trenbolone Acetate Metabolites in Simulated Confined Animal
- Feeding Operation (CAFO) Runoff. *Environmental Science & Technology* 2012, 46, (7), 38033810.
- 297 3. Schwarzenbach, R. P.; Gschwend, P. M.; Imboden, D. M., *Environmental organic chemistry*. 2nd ed.; Wiley-Interscience: New York, N.Y., 2003; p xiii, 1313 p.
- 299 4. Othmer, D. F.; Thakar, M. S., Correlating diffusion coefficient in liquids. *Industrial & Engineering Chemistry* 1953, 45, (3), 589-593.
- 301 5. Wilke, C.; Chang, P., Correlation of diffusion coefficients in dilute solutions. *AIChE*302 *Journal* 1955, *1*, (2), 264-270.
- 303 6. Hayduk, W.; Laudie, H., Prediction of diffusion-coefficients for nonelectrolytes in dilute
  aqueous-solutions. *Aiche Journal* 1974, 20, (3), 611-615.
- 305 7. Millington, R.; Quirk, J., Permeability of porous solids. *Transactions of the Faraday*306 *Society* 1961, *57*, 1200-1207.
- 8. Khan, B.; Qiao, X.; Lee, L. S., Stereoselective sorption by agricultural soils and liquid-
- **308** liquid partitioning of trenbolone (17 $\alpha$  and 17 $\beta$ ) and trendione. *Environmental Science* & **309** *Technology* **2000** *42* (22) 8827 8822
- 309 *Technology* 2009, *43*, (23), 8827-8833.
  310