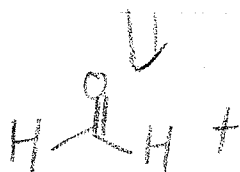
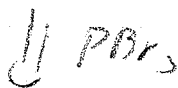
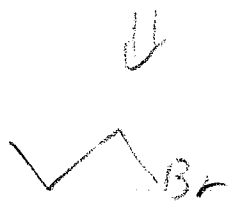
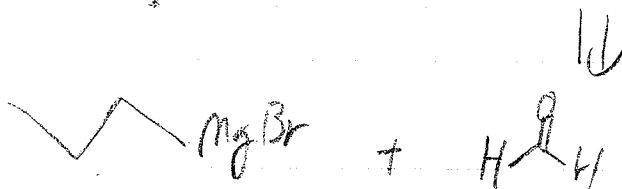
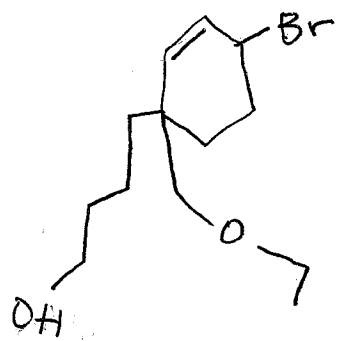


Good!!
All of

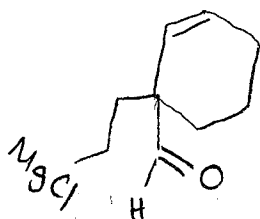


Ashley Morgan

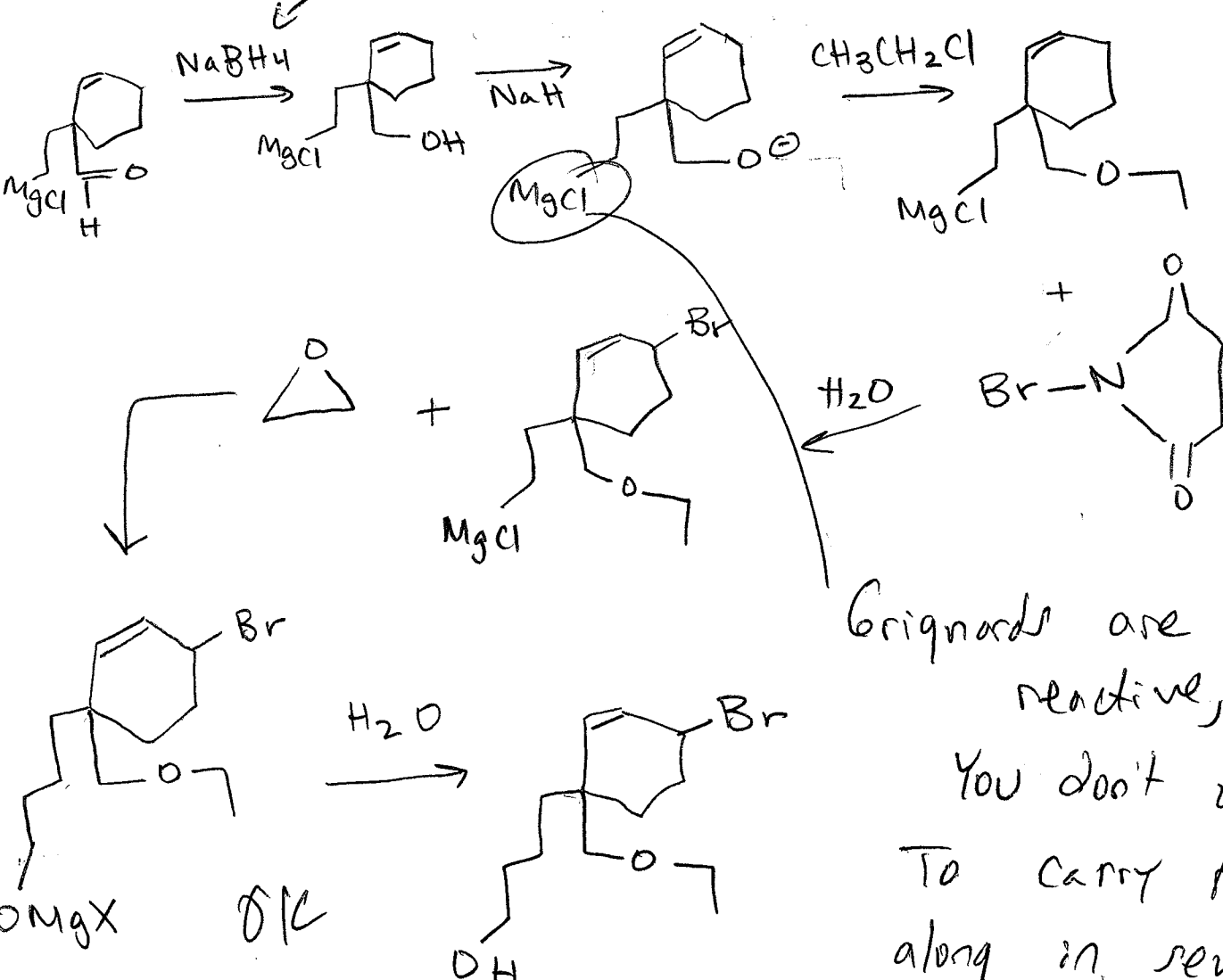
Synthesize this molecule:



From this starting material:



Answer:

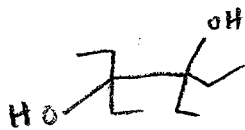


it will survive reagent here

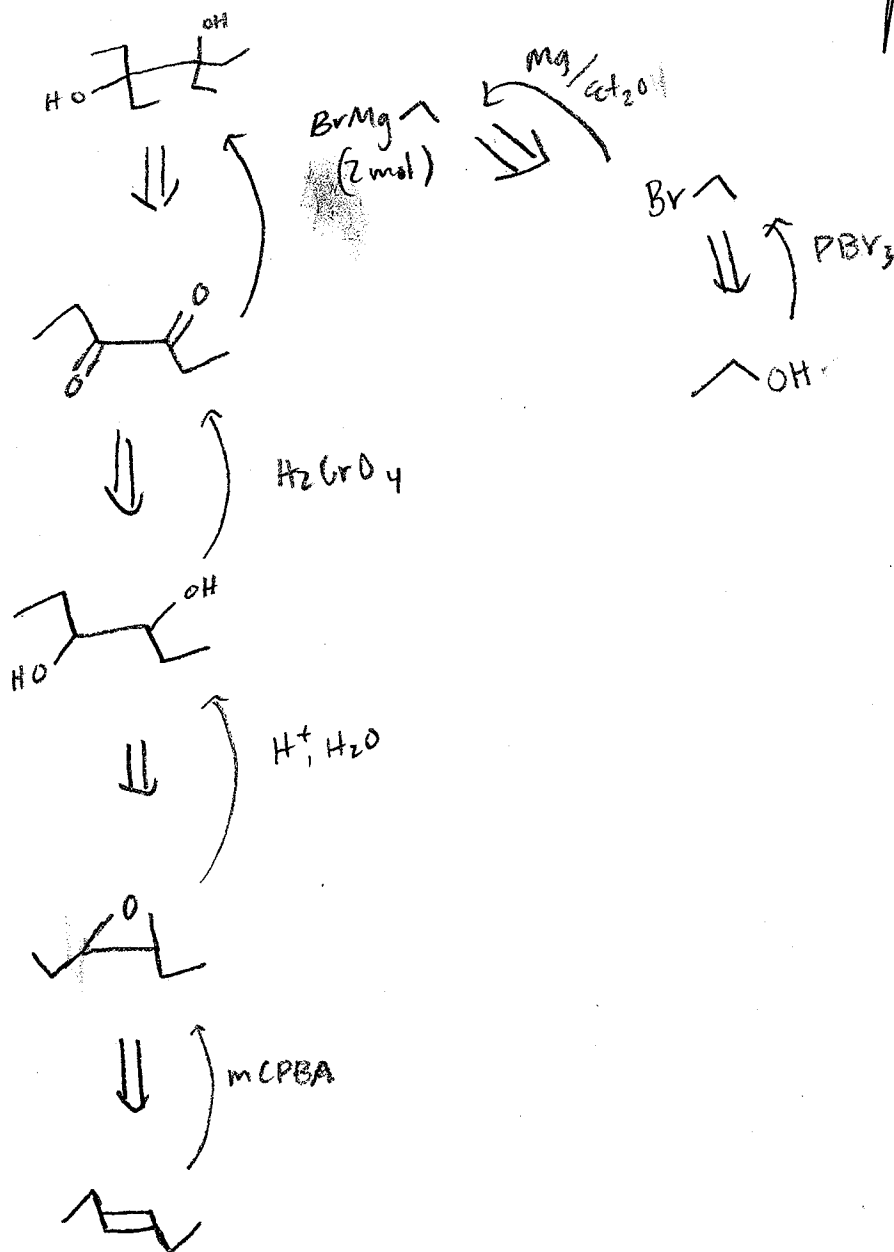
I'm not sure w/ Grignard

Grignards are reactive, you don't want to carry them along in several steps.

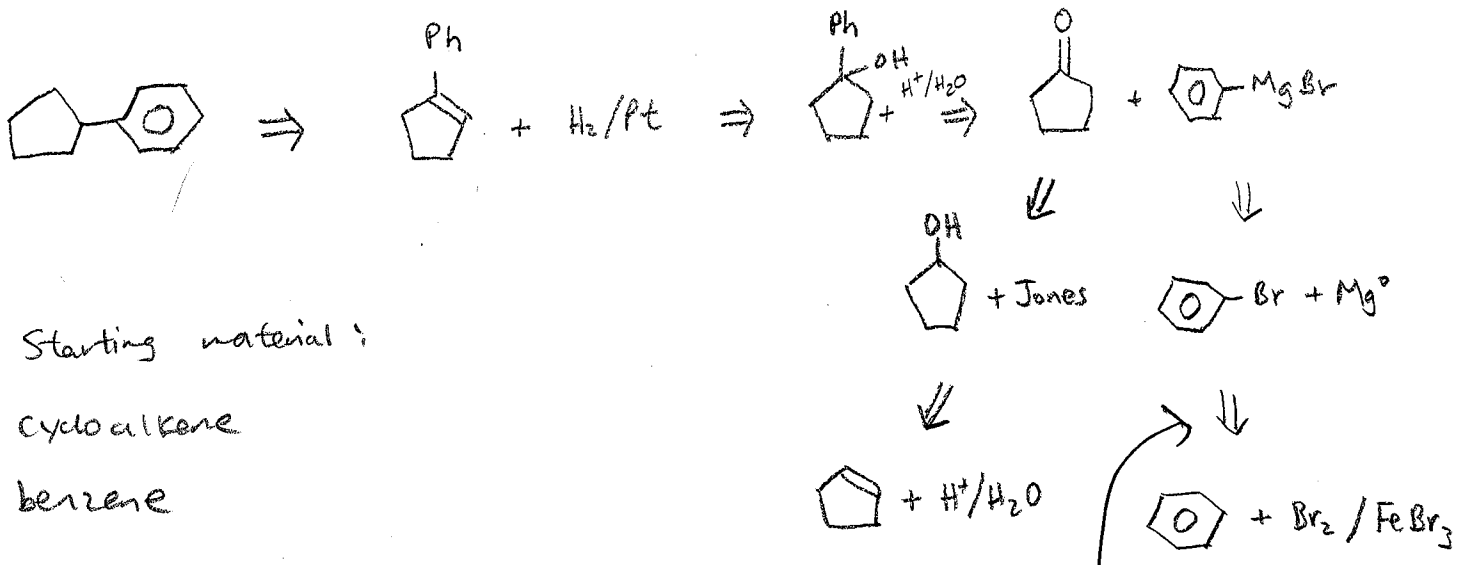
synthesize the following molecule:



Answer



Good !!
MG



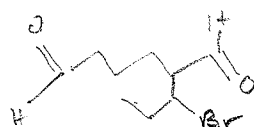
Starting material:
 cycloalkene
 benzene

good !!
 MB

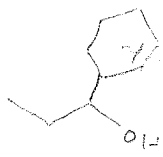
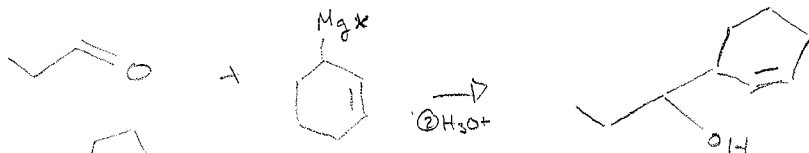
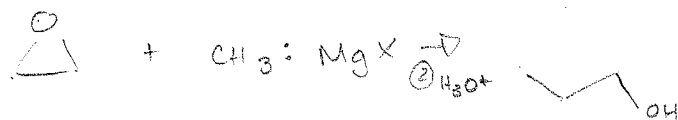
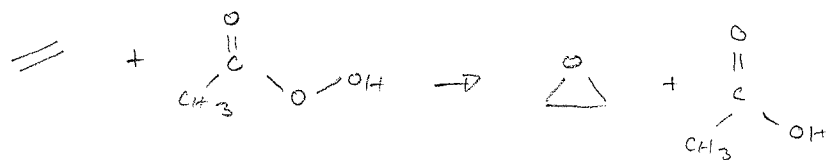
David Zhu
 chem 336 AB
 2/8/05

ok will
 cover
 for
 in
 next
 chapters

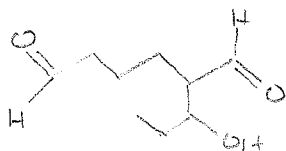
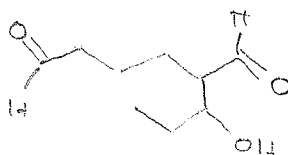
Start with $=$ to make more than 3 carbons.



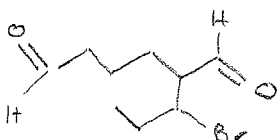
Reagents may be



① O_3
② Zn/HOAc

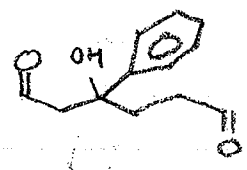



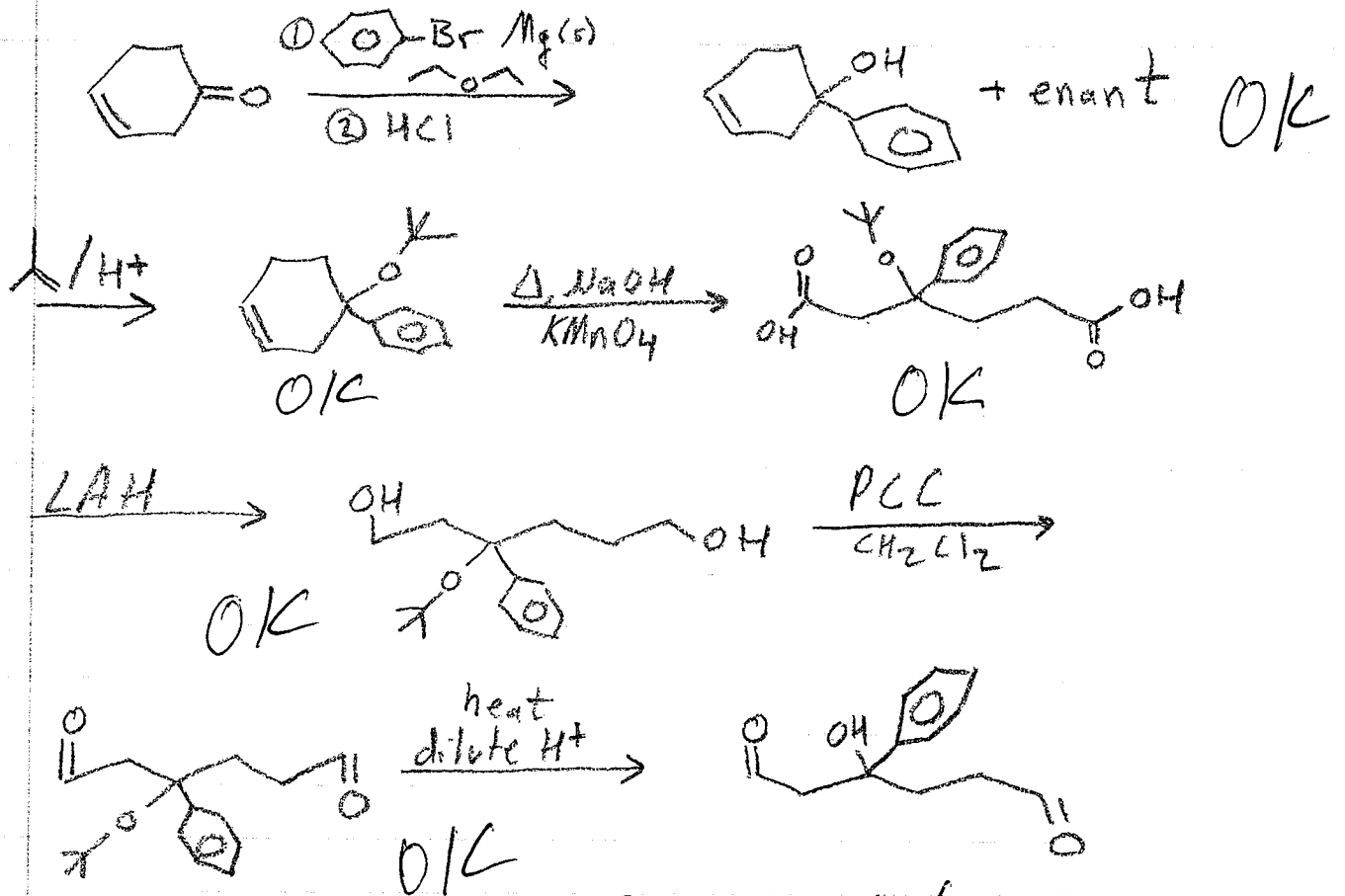
PBr_3



good !!!
MF

Shane Kinard
218105
CHEM 336
Synthesis 3

Synthesize  as a racemate from 
and any other necessary reagents.

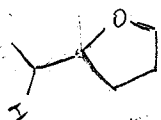


Good !!
Mb

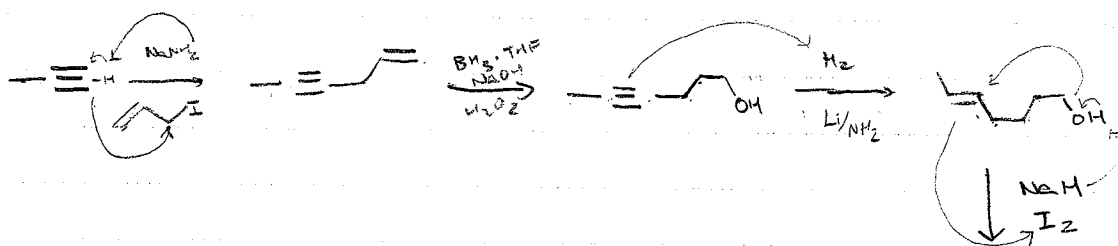
Tim Mc

CHEM 336

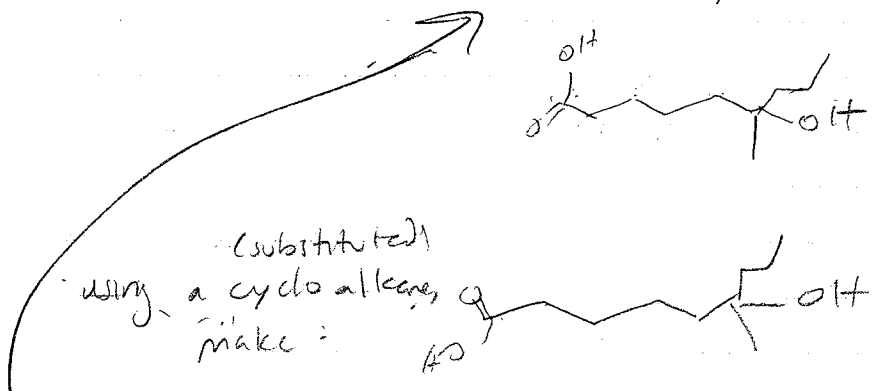
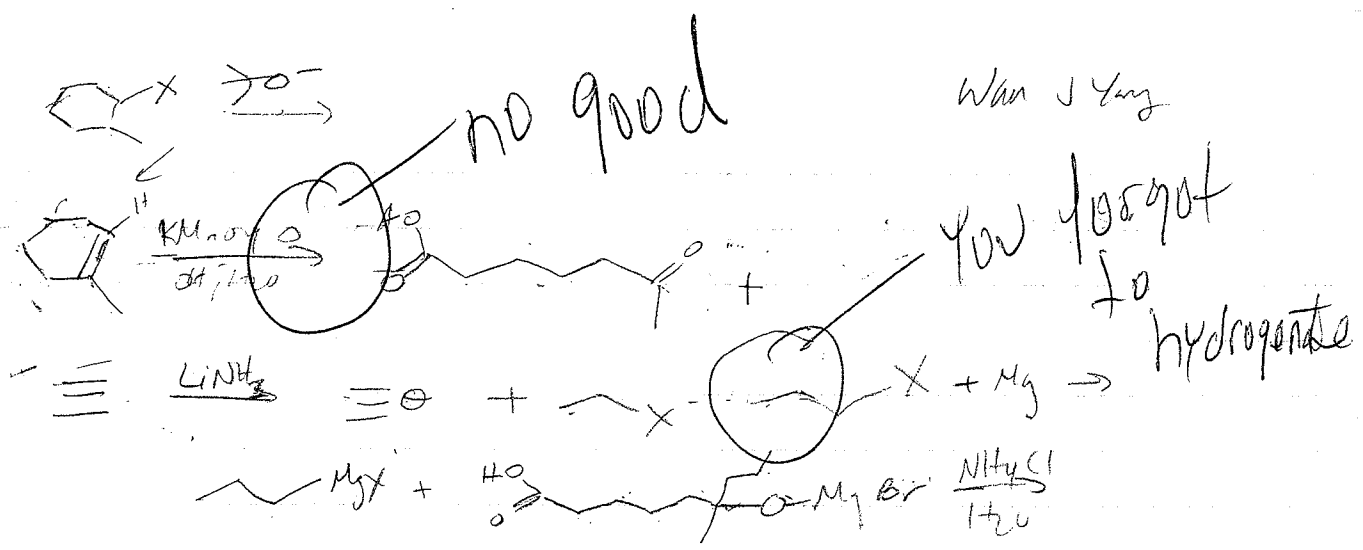
MECHANISM HOMEWORK



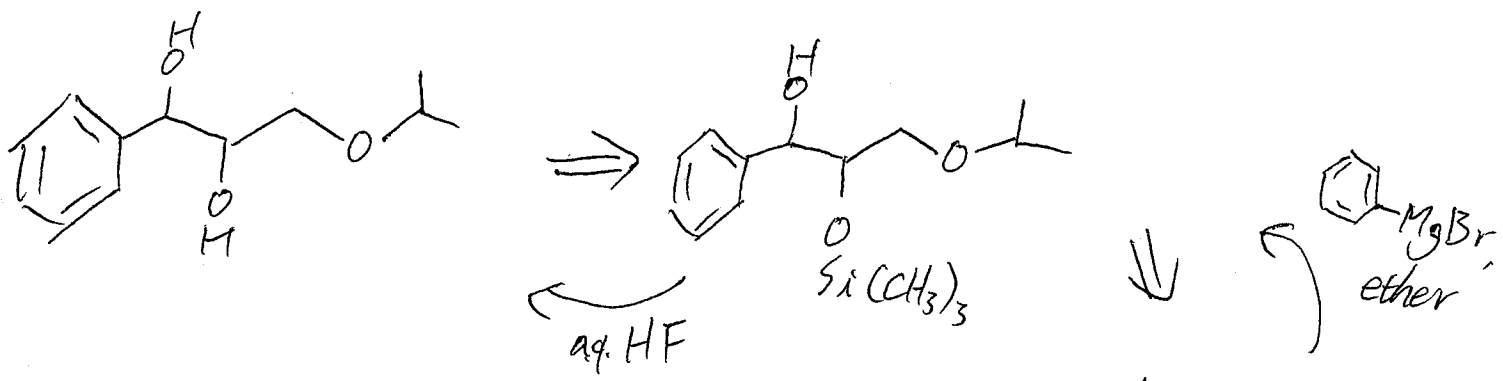
less than 4 carbon - (1, 2, 3)



where
did you
see this
rxn?

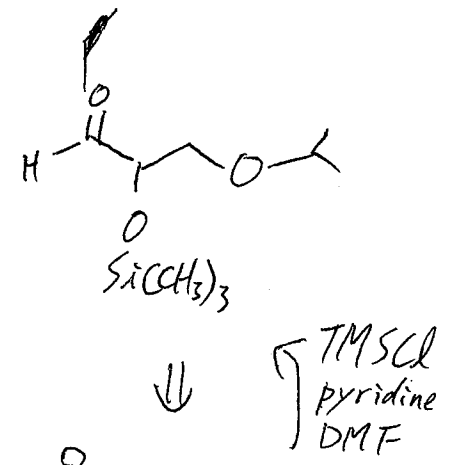


grignard also will react w/ COOH end.

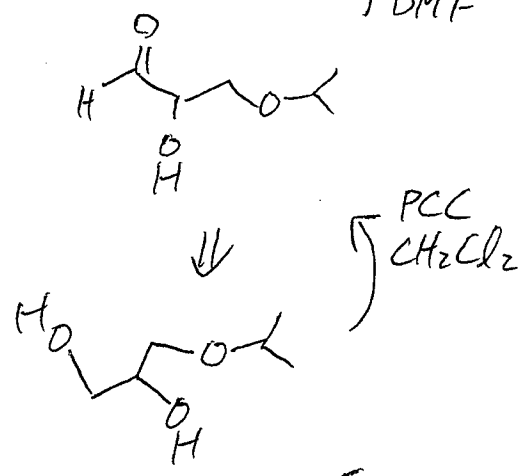


Starting material.

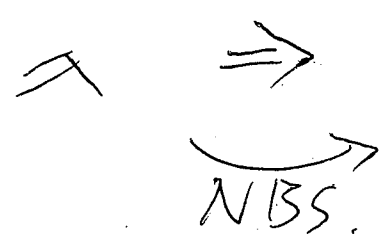
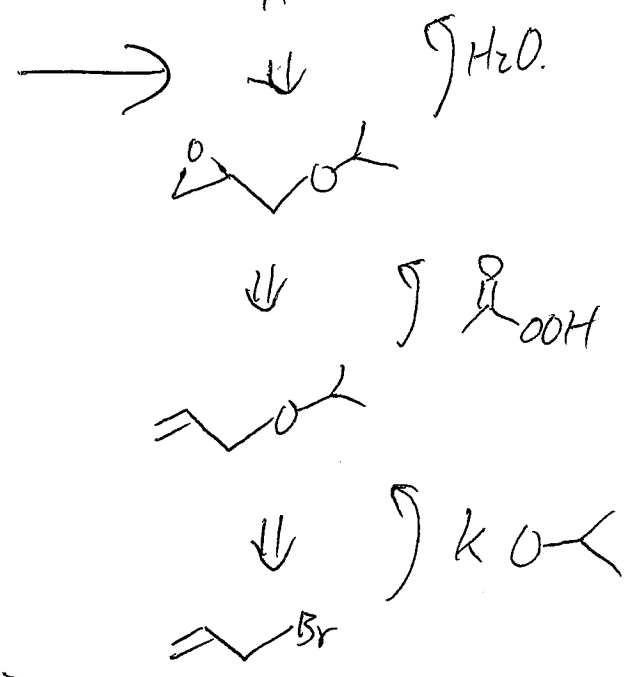
3 carbons or less.
(or propene)



pcc will also oxidize II R₂H to ketone



need some OH⁻ or H⁺ w/ the H₂O



NOT too bad

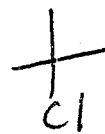


Hsin-yi Lin

A!

- ① Start w/ $\text{CH}_3\text{CH}_2\text{CH}_3$
- ② $\text{CH}_3\overset{\text{CH}_3}{\text{CH}}\text{CH}_3 + \text{Cl}_2 \xrightarrow{\text{light}} \text{CH}_3\overset{\text{CH}_3}{\text{CH}}\text{CH}_2\text{Cl}$
- ③ $\text{CH}_3\overset{\text{CH}_3}{\text{CH}}\text{CH}_2\text{Cl} + \text{OH}^- \rightarrow \text{CH}_3\overset{\text{CH}_3}{\text{CH}}\text{CH}_2\text{OH}$
- ④ $\text{CH}_3\overset{\text{CH}_3}{\text{CH}}\text{CH}=\text{CH}_2 + \text{H}_2\text{SO}_4 \rightarrow \text{CH}_3\overset{\text{CH}_3}{\text{C}}=\text{CH}_2$
- ⑤ $\text{CH}_3\overset{\text{CH}_3}{\text{C}}=\text{CH}_2 + \text{KMnO}_4 \rightarrow \text{CH}_3\overset{\text{CH}_3}{\text{C}}=\text{O} + \text{O}=\text{C}=\text{O} + \text{H}_2\text{O}$

but would also get

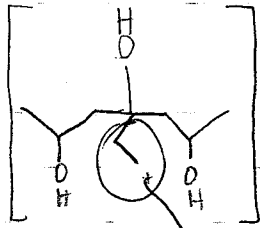


- ⑥ $\text{CH}_3\text{CH}_3 + \text{Br}_2 \xrightarrow{\text{light}} \text{CH}_3\text{CH}_2\text{Br}$
- ⑦ $\text{CH}_3\text{CH}_2\text{Br} + \text{Mg} \xrightarrow{\text{ether}} \text{CH}_3\text{CH}_2\text{MgBr}$
- ⑧ $\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_3 + \text{CH}_3\text{CH}_2\text{MgBr} \rightarrow \text{CH}_3-\overset{\text{CH}_3}{\underset{\text{OH}}{\text{C}}}-\text{CH}_2\text{CH}_3$) OK

good MB

Question! Synthesize $\text{CH}_3-\overset{\text{CH}_3}{\underset{\text{OH}}{\text{C}}}-\text{CH}_2\text{CH}_3$ from compounds w/ 3 carbons or less.

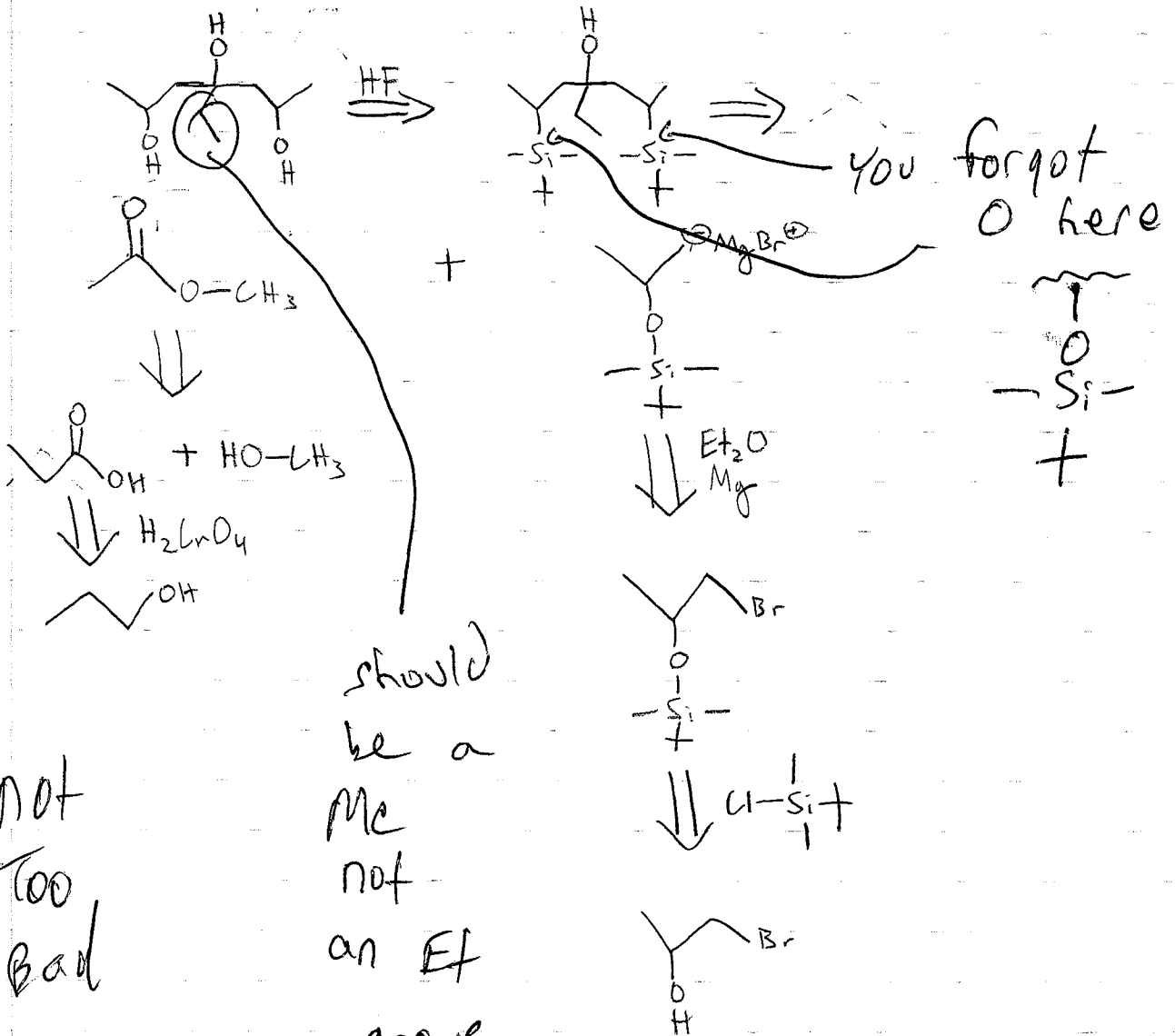
Make



Using 3 carbon or less alcohols and alkyl halides using any reagents.

should be Me not Et

Answer: (Retrosynthesis)



not too bad Mb

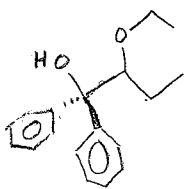
should be a Me not an Et group

Tri Le

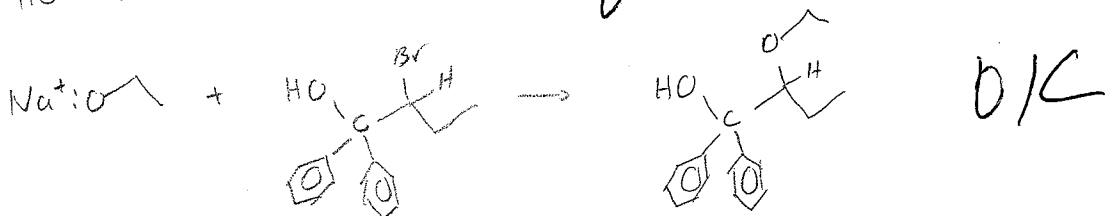
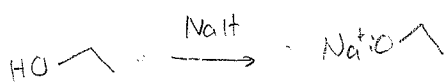
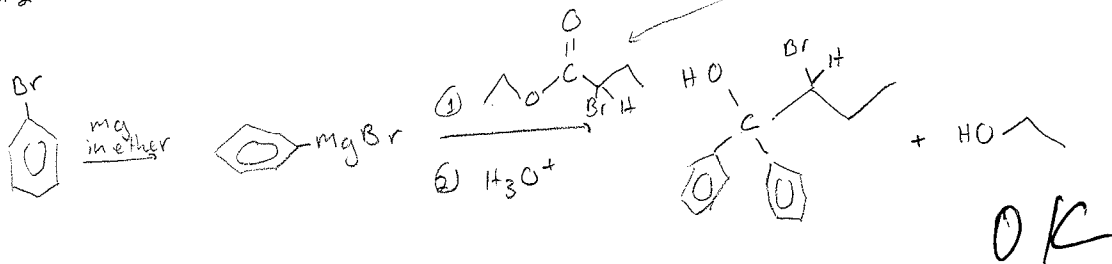
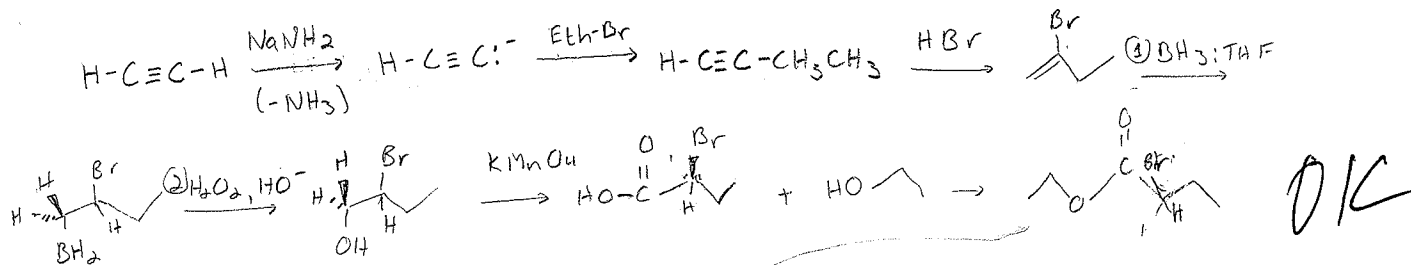
Chem 336 AA

Synthesis problem

start w/ 2 carbon and benzene.



is this OK?

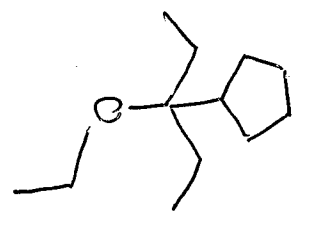


looks good

can except perhaps 1 questionable step.

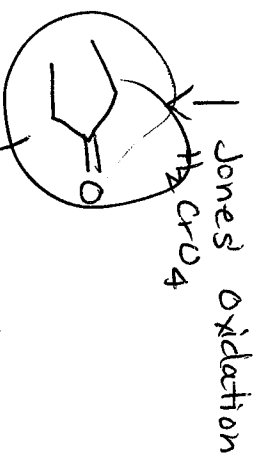
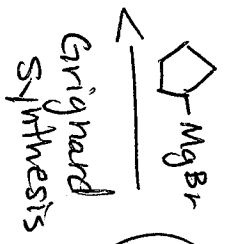
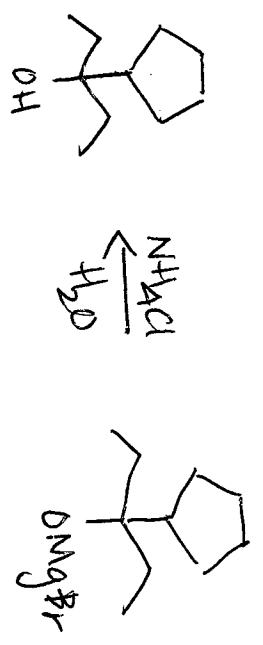
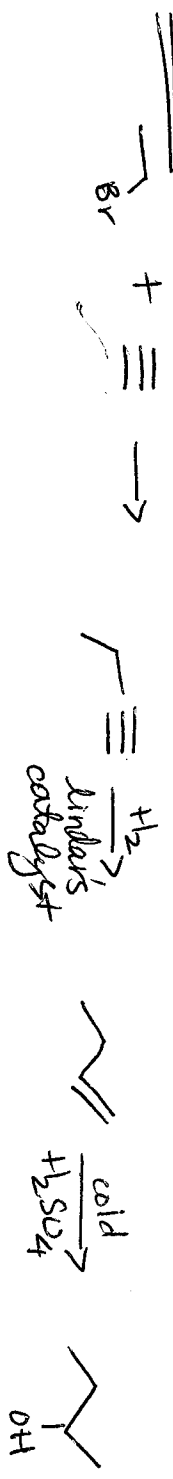
MF

- I like to call this my "one-leg-pirate" (make it out of 2 carbons (or less))
 *I don't know how to give him a peg.

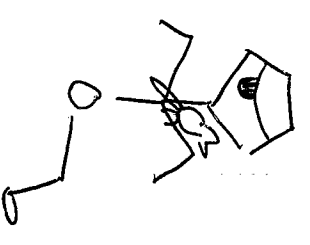


yes!
 Mr

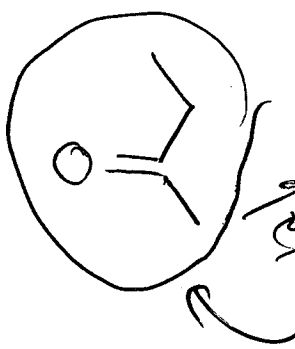
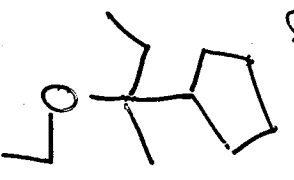
Answer:



Williamson Ether Synthesis



so you actually can make

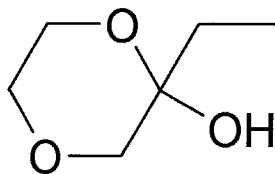


no, you can't

which is a one-leg, one-arm pirate !!!

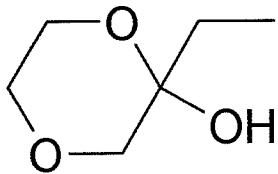
Name: _____

(points) Propose a synthesis of the following compound from starting materials containing 2 carbons or less and no oxygen atoms and any other needed reagents. For each of the necessary reactions, indicate the reagents needed. Note, there is no restriction on the number of carbons and oxygen in the reagents. Be sure to circle your final synthesis sequence so that the TA can grade it.



Name: _____

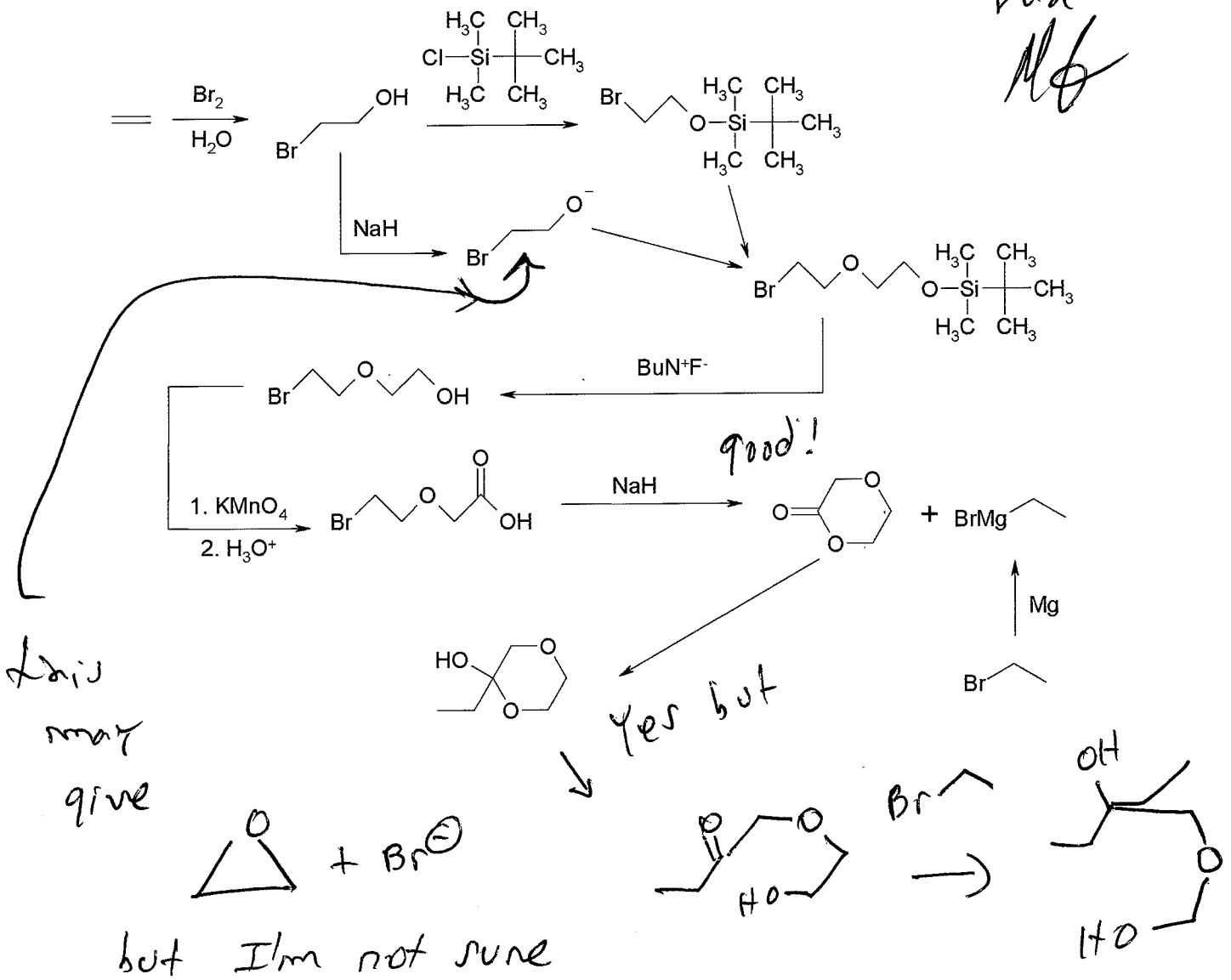
() points) Propose a synthesis of the following compound from starting materials containing 2 carbons or less and no oxygen atoms and any other needed reagents. For each of the necessary reactions, indicate the reagents needed. Note, there is no restriction on the number of carbons and oxygen in the reagents. Be sure to circle your final synthesis sequence so that the TA can grade it.



Wow Typed
Yah!

Answer

not too
Bad
MB



Synthesis HW

Joshua Baguch

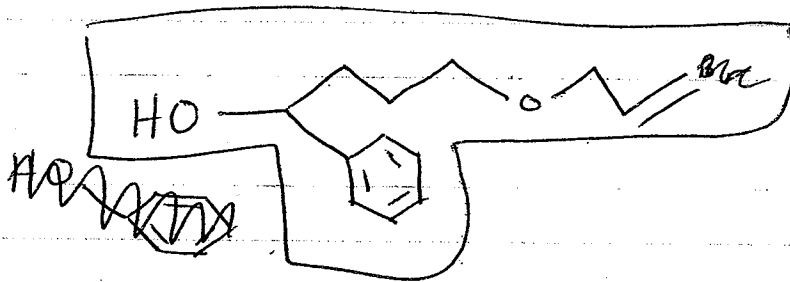
Chem 346

2/9/05

Starting with CH2=CHCHO and

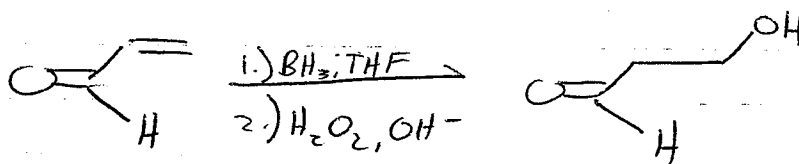
using any reagents necessary,

synthesize the following compound:

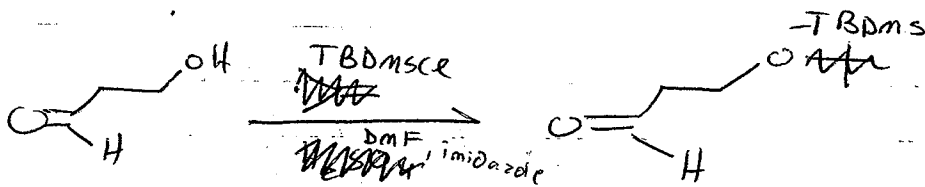


Answer

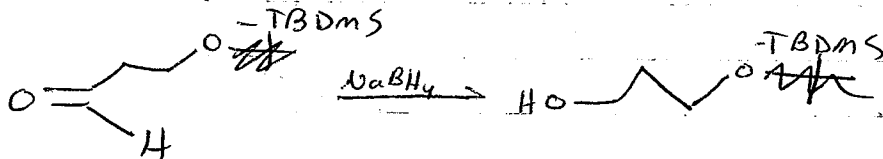
(1)



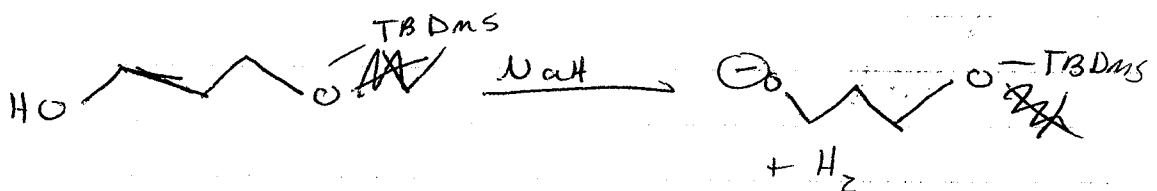
(2)



(3)

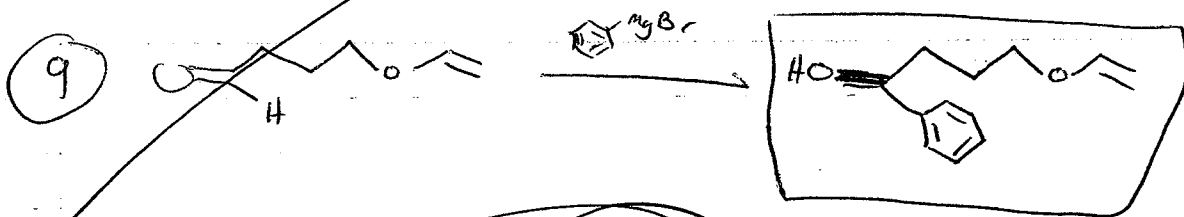
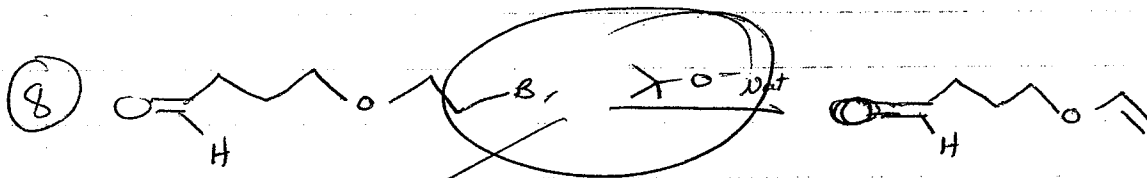
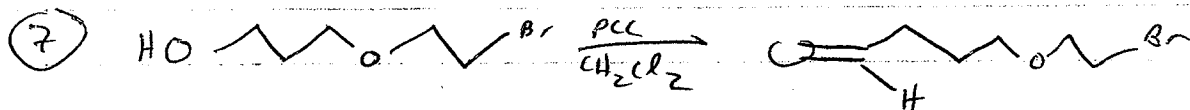
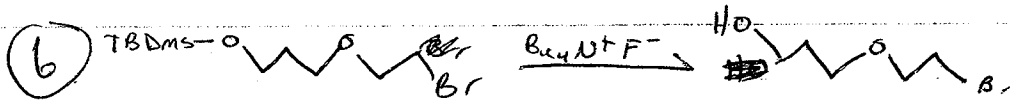
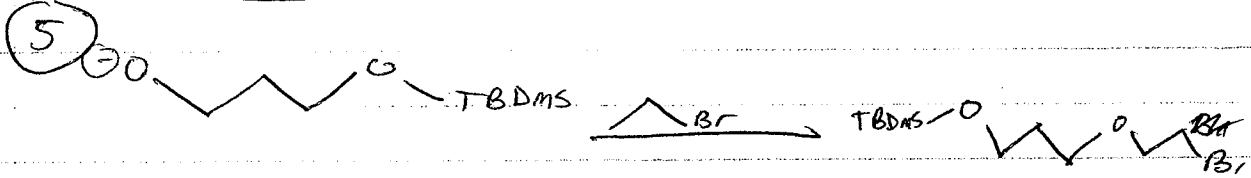


(4)



(over)

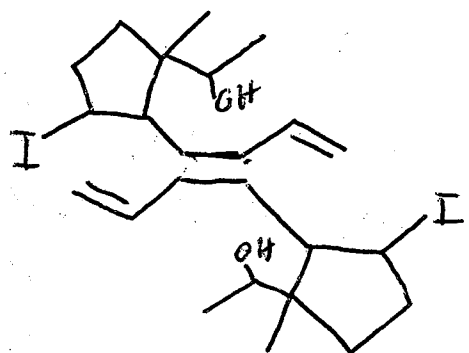
Answer (cont.)



not
sure
about
this
step

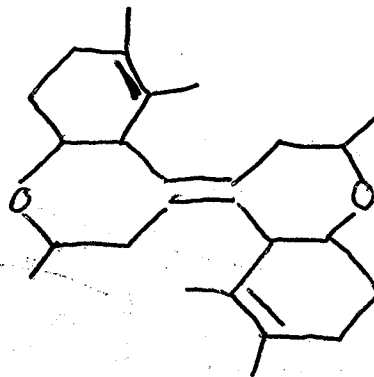
good Me

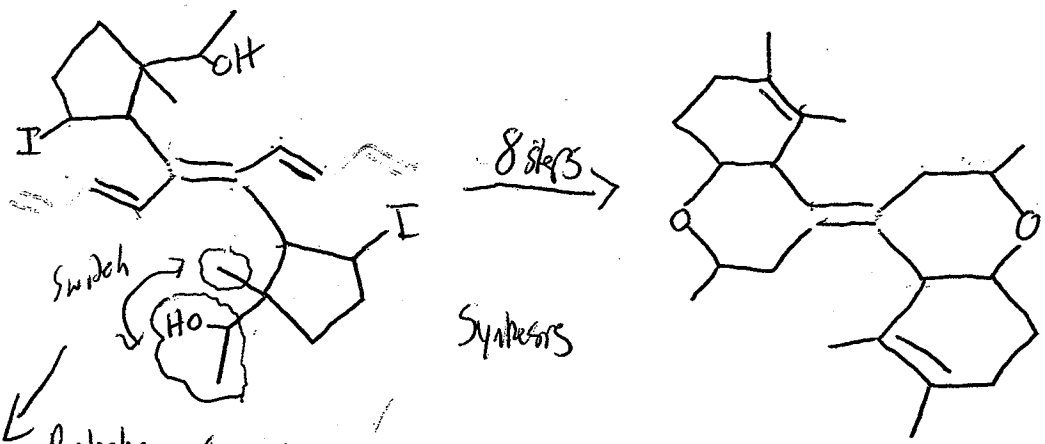
Synthesis



8 steps
⇒

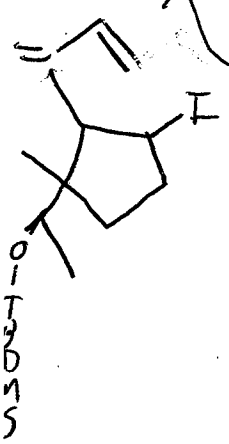
Final product



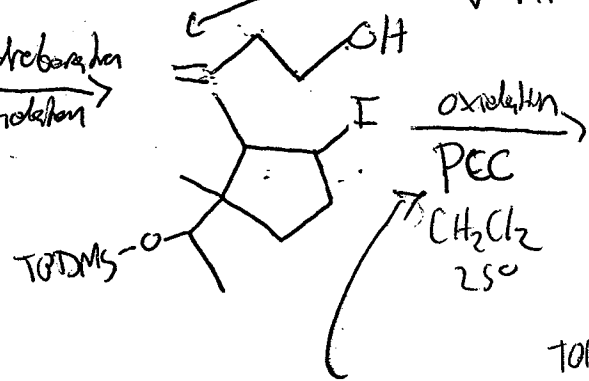


Protecting group (TBDMS) DMF

what is dms step?
dms may also react w/ BH_3

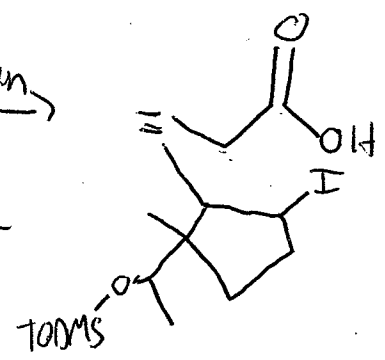


Hydroboration
Oxidation



oxidation

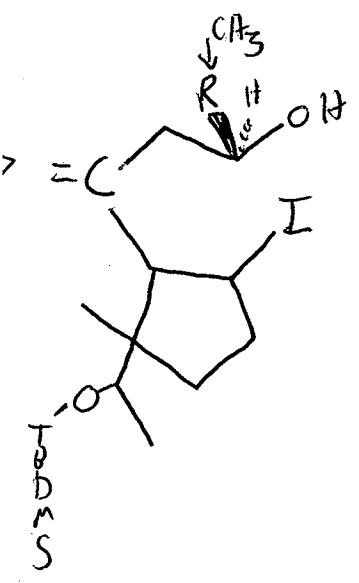
PEC
 CH_2Cl_2
 25°



Grignard
 CH_3MgX

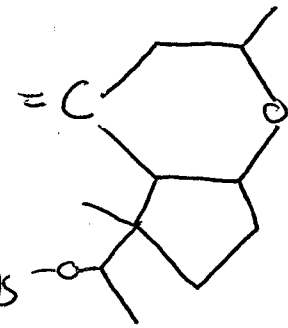
NO
gives III
ROH not II

gives $R-C(=O)H$ not $R-COOH$

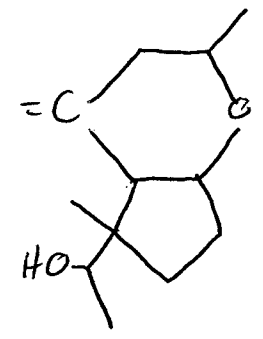


NaH
Willkinson
Synthesis

good



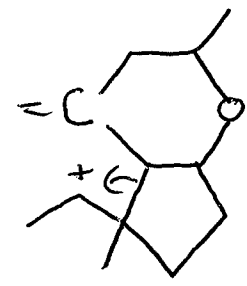
F^-



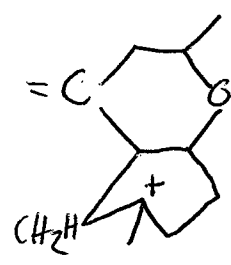
OK

Carbocation rearrangement

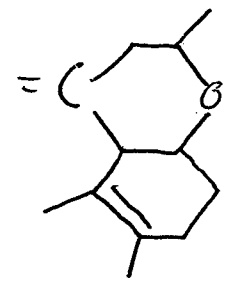
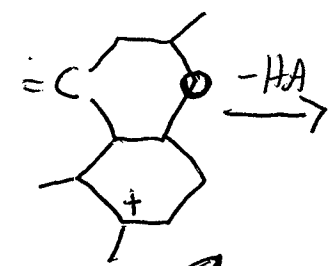
HA
end
-H₂O



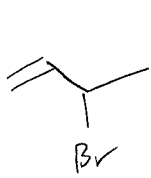
OK



I think

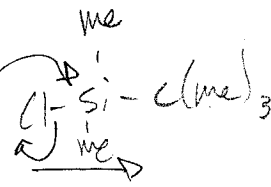
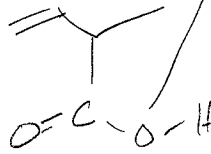


Tyler Swager

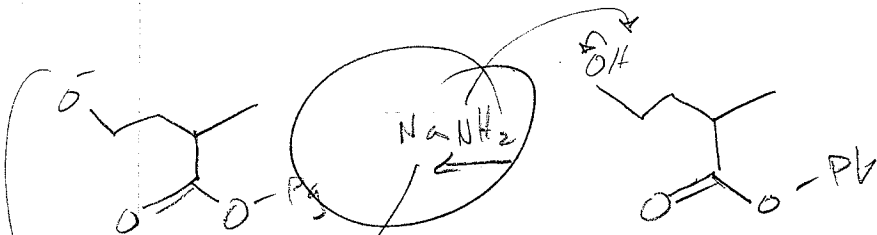
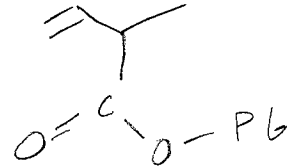


(1) Mg(s) in THF
 (2) O=C=O
 (3) H₂O

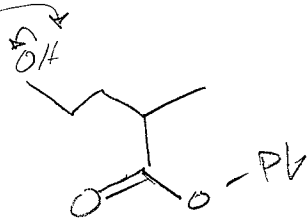
OK



OK

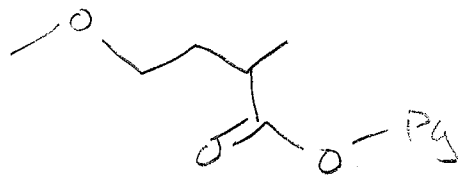


NaNH₂



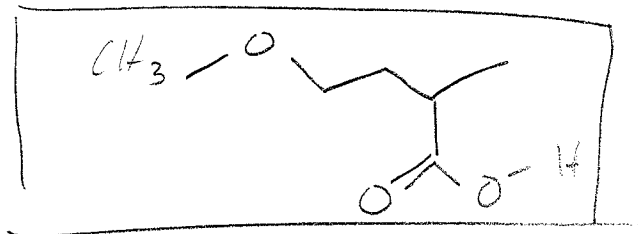
BH₃THF
 OH⁻, H₂O₂

OK



Bu₄N⁺F⁻

Start with 4 carbons
 or less. A halide will
 be present.

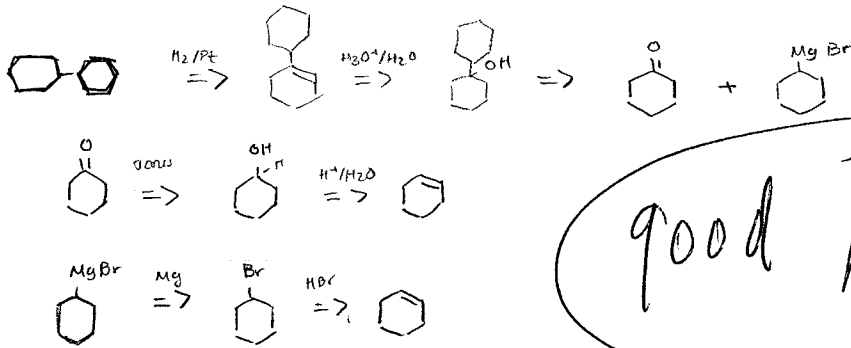


great!

MB

OK

~~Want to see other~~
~~but~~



good Mg

Make prof's Gelb's glasses with a cycloalkene of 6 carbons or less.

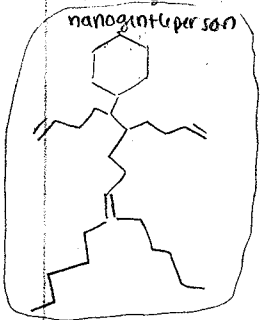


not good Mg

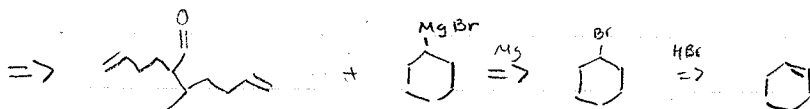
Protect society!

Take away nanopirate's alcohol and make him a respectable

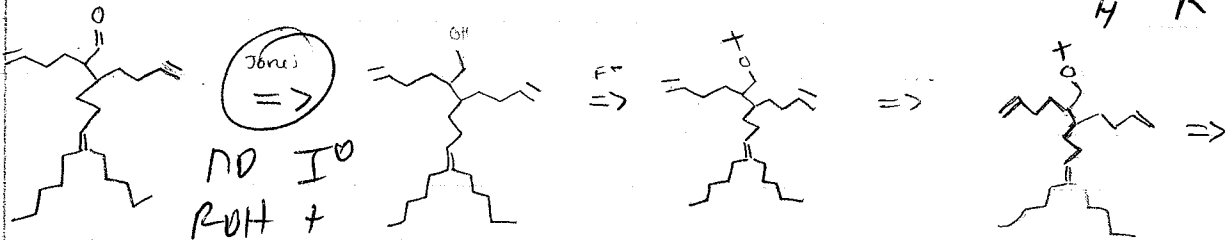
Nanogentleperson! (AKA Rehabilitation by oxidation) (by Laura Walsh and Jammie Yang)



(oxobot)

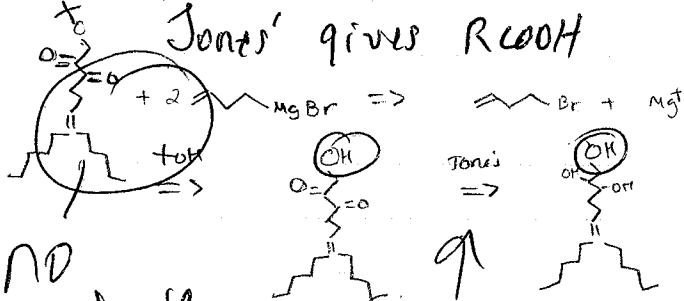


\uparrow NO $R-C(=O)H + R'MgBr$
 gives $R-C(OH)(H)-R'$



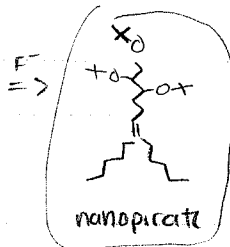
NO IO
 ROH +

Jones' gives RCOOH

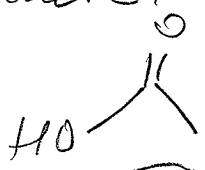
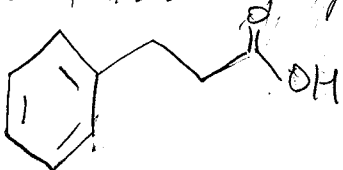


NO
 as above

NO
 as above



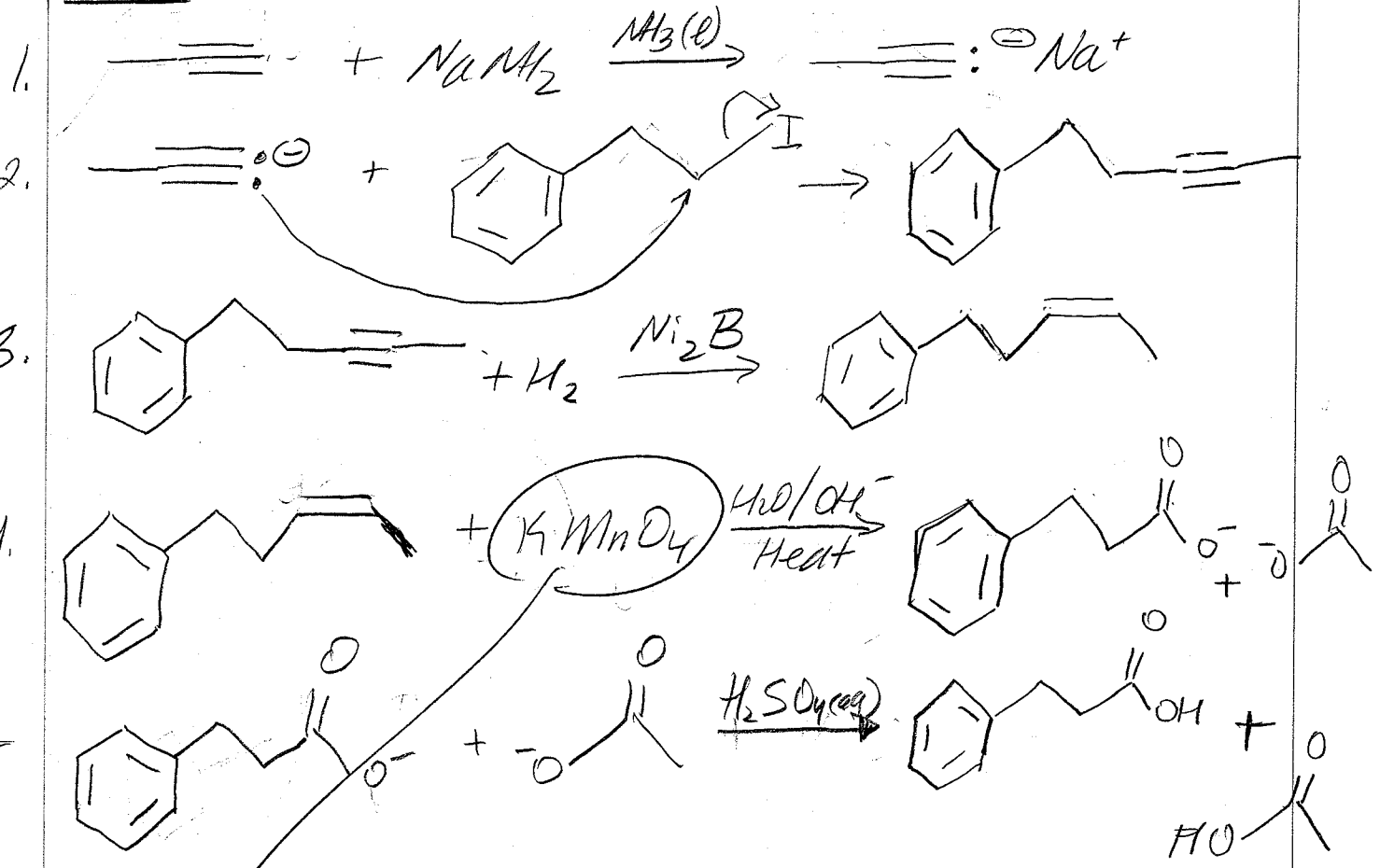
Using 1-propyne and 1-iodo-2-phenylethane reagents, give a practical synthesis of the following products:



Answer:

22-141 50 SHEETS
22-142 100 SHEETS
22-144 200 SHEETS

CAMPAL

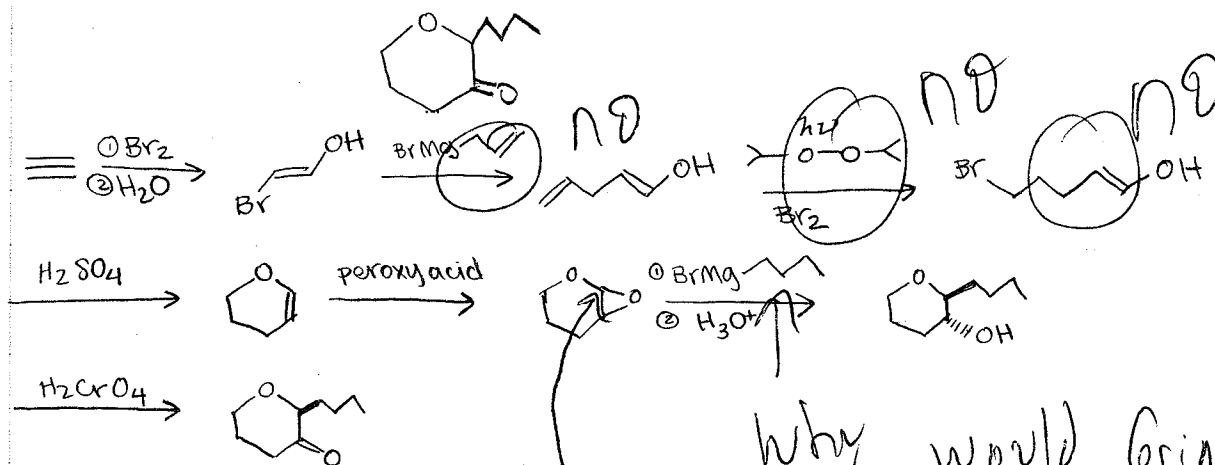


Good

I think this reaction will give [O]C(=O)C1=CC=CC=C1 but this will be shown in a later chapter

Rebecca Reh

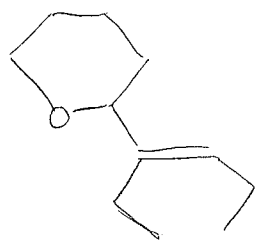
Synthesis the following compound beginning with a 2 carbon alkyne.



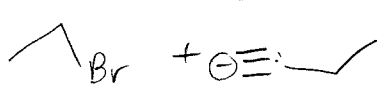
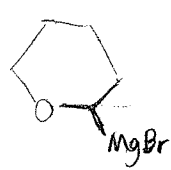
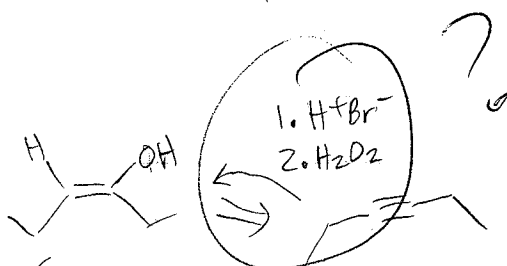
Why would Grignard go to only only

donis carbon ?

3 C or less

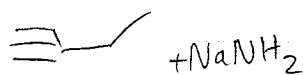


⇓

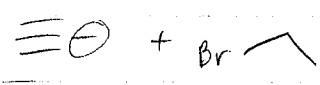


⇓

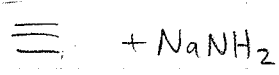
⇓



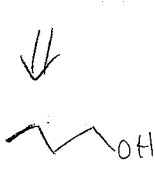
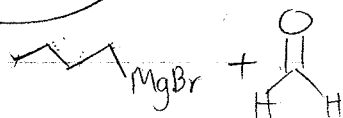
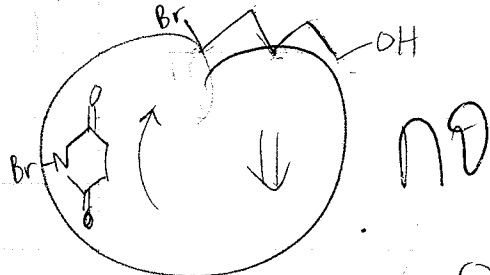
⇓



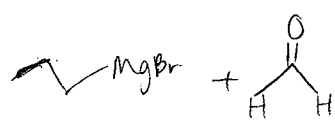
⇓



1. NH_3
2. Br-N
3. Grignard



⇓



⇓

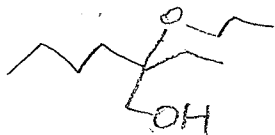


Grignard

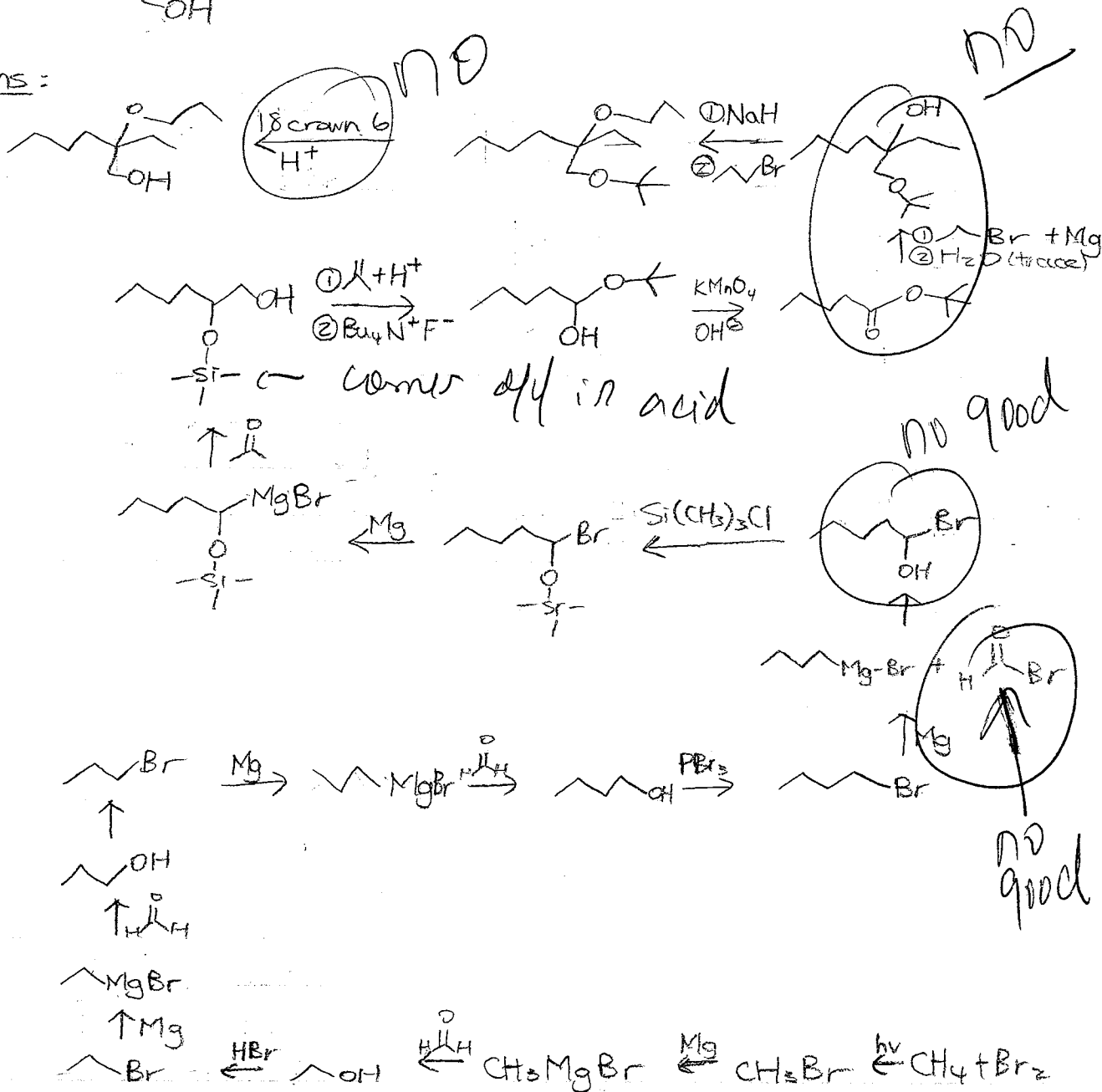
Timothy Chao
2-7-05

Chem 336
Retrosynthesis

- ① Device a synthesis of the following compound starting compounds containing One carbon:

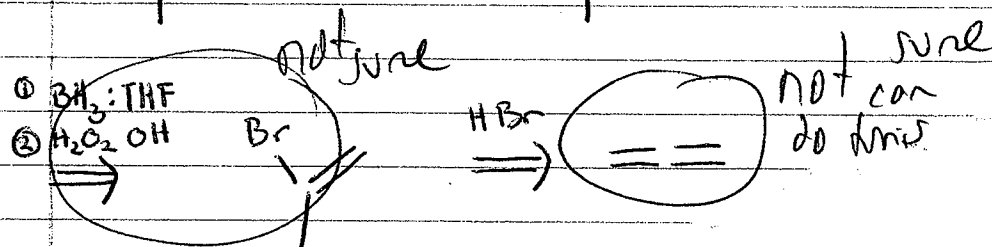
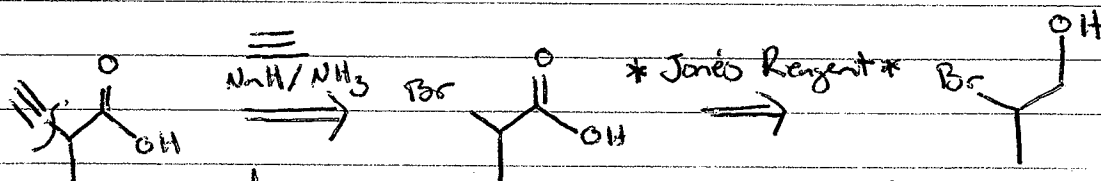
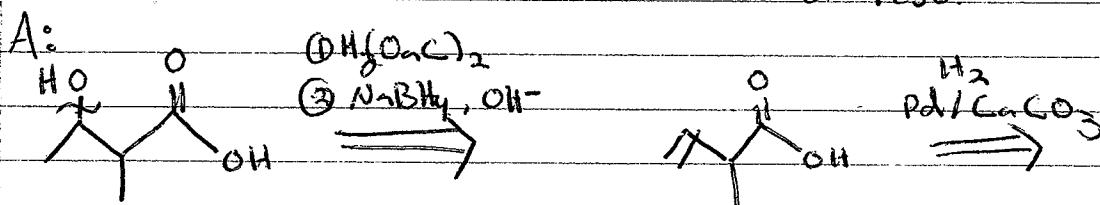


Ans:

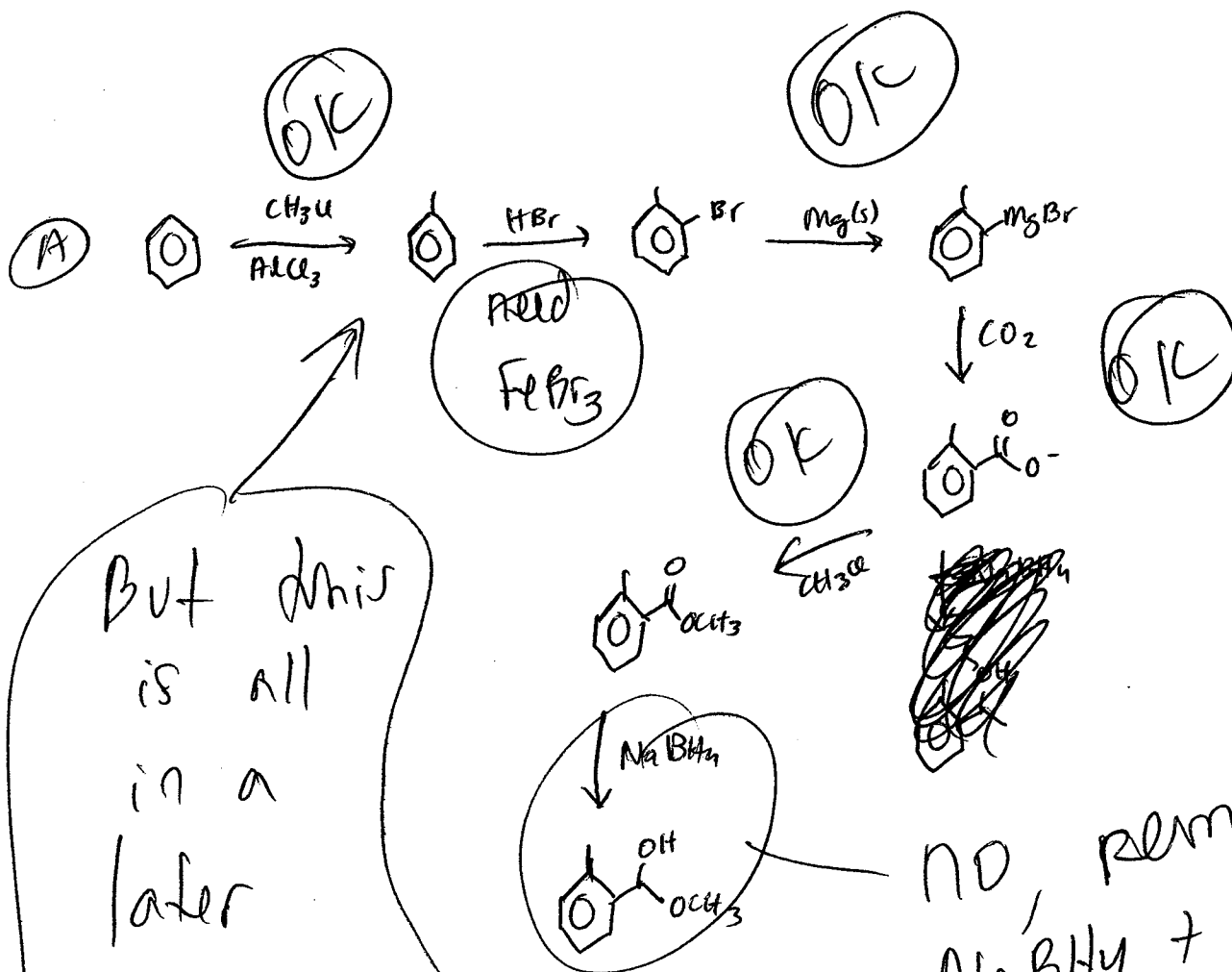
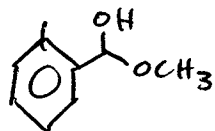


Audrey M.
2/9/05

Q: Make CC(O)C(=O)O from a starting compound consisting only of 3 carbons or less.



⇒ Starting with benzene & using no other starting materials with >1 carbon, synthesize:

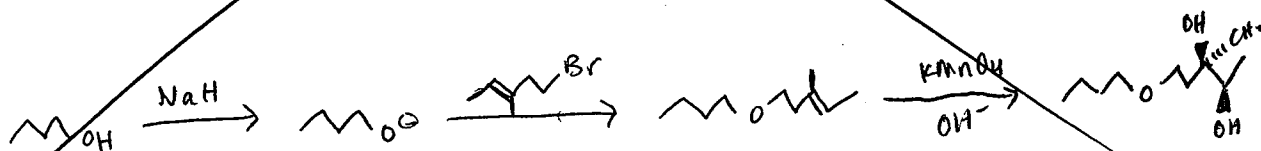
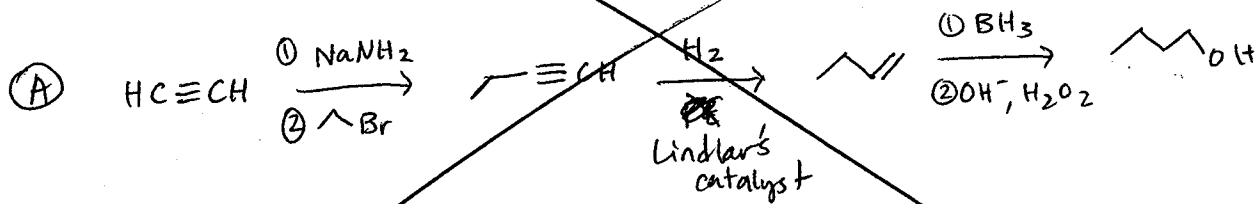
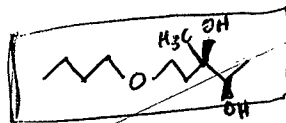


But this is all in a later chapter

NO, remember NaBH₄ + ester → no rxn.

That we will come to soon.

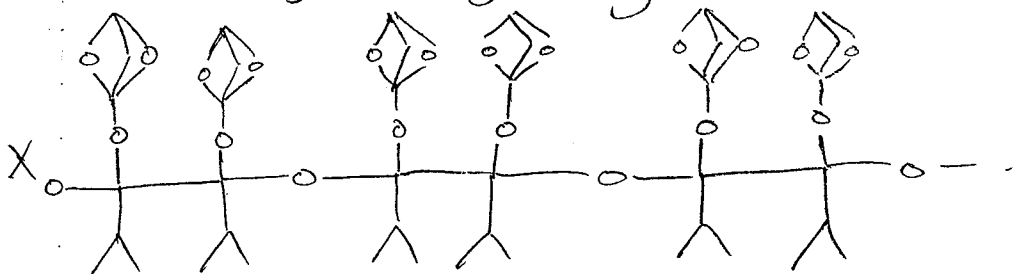
⇒ No starting materials with >2 carbons. Synthesize



Disclaimer: I have no idea if this will work & I certainly don't think it would be a major product. But it looked so silly, not to try it.

Nick Cox

Seeing as this exam will be on Valentine's Day, I figured I'd use the holiday as sort of a theme for my synthesis. The end goal is to prepare this polymer using starting compounds of 3 carbons or less:

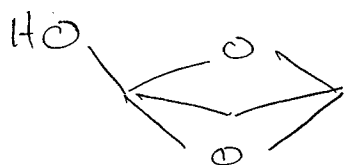


Overall there is some valid & invalid chemistry here

I suppose that since I'm a member of the ACLU, avid supporter of same-sex marriages, and since it makes this synthesis a lot easier, neither of the two anthropomorphic groups have my gender defining features.

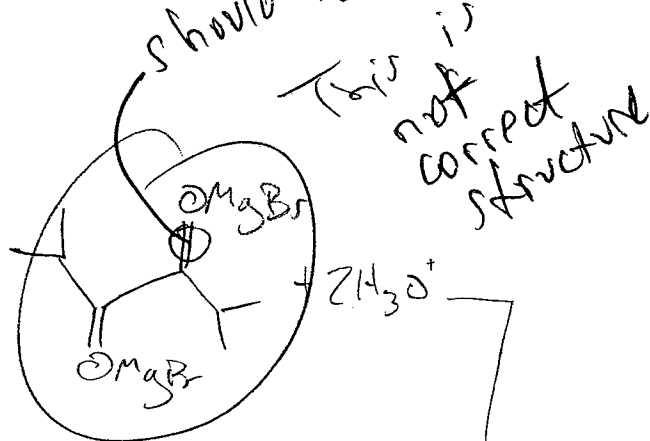
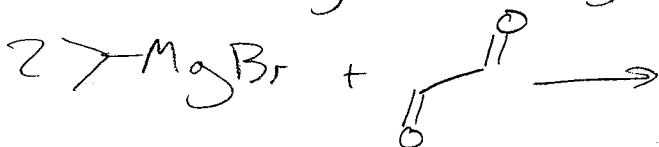
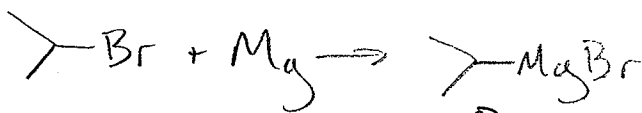
Mo

We need to make "legs" and "heads". Heads are easy since



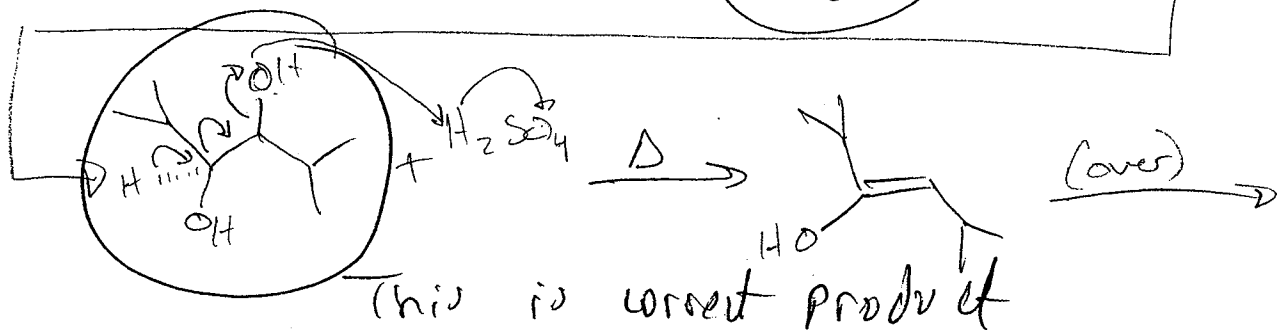
only has 3 carbons.

Legs on the other hand...

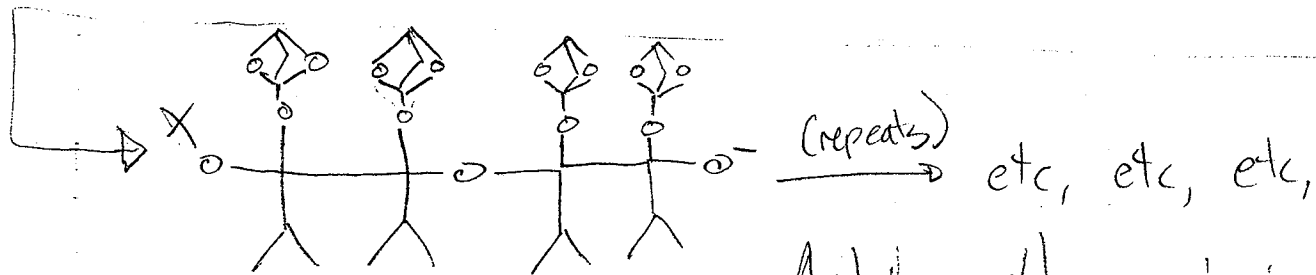
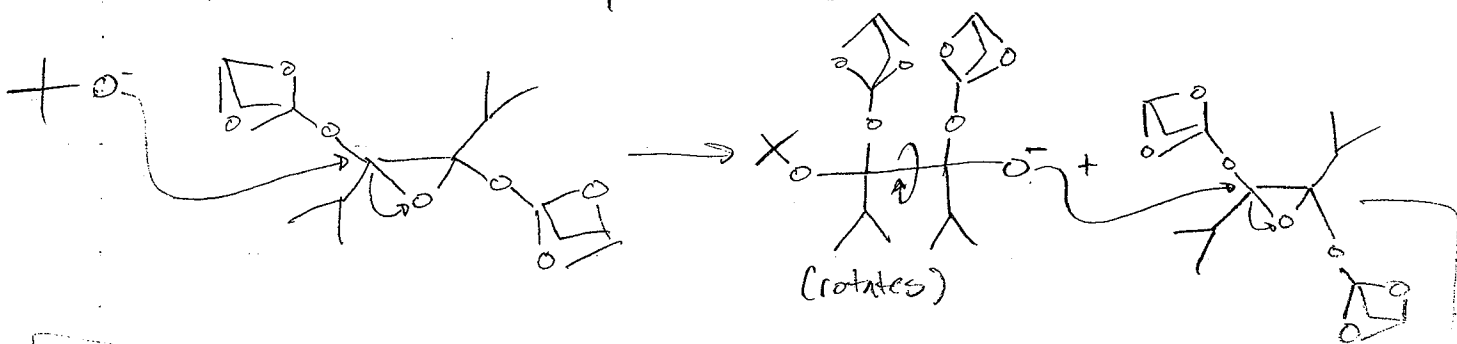
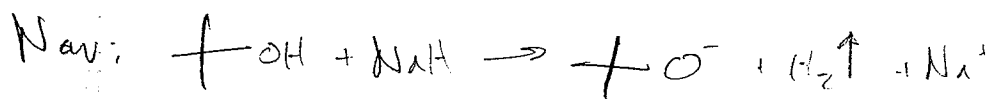
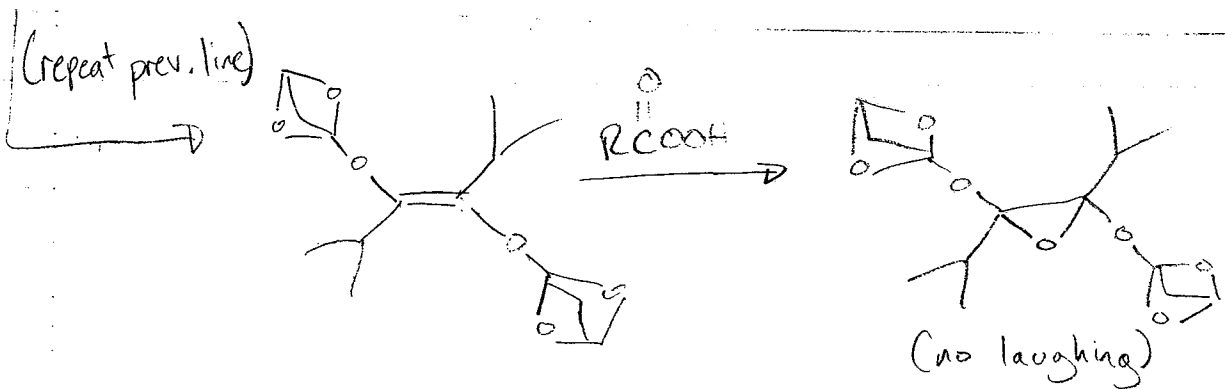
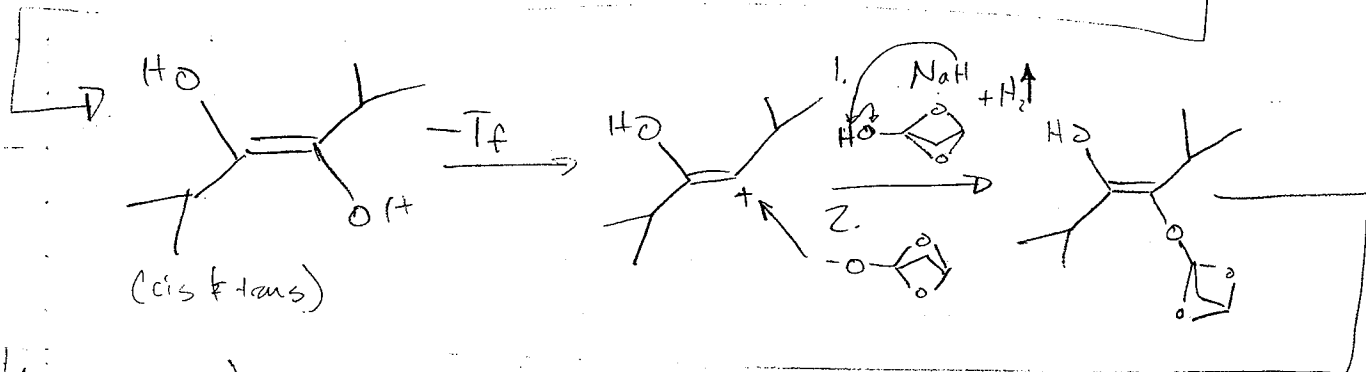
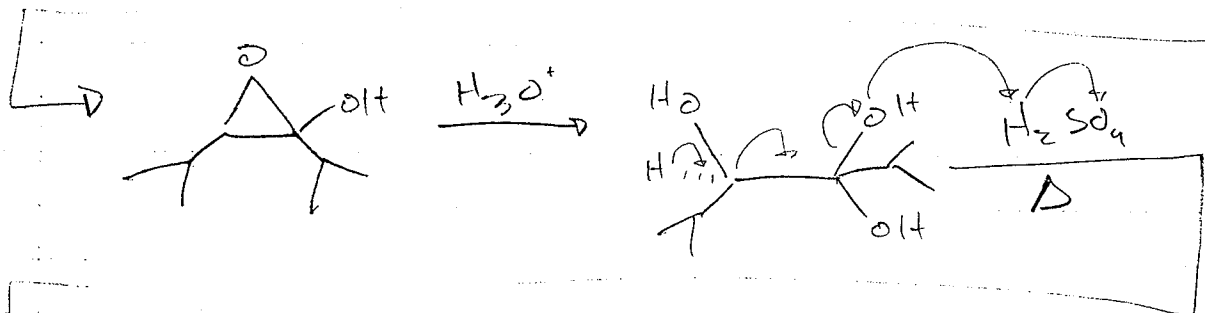
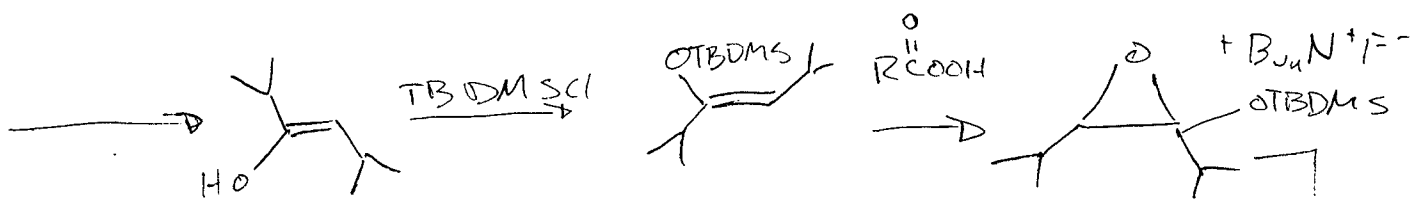


should be single bond

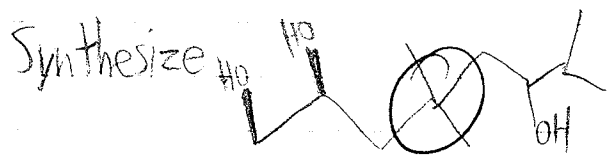
This is not correct structure



(over)



And the world was a happier place.

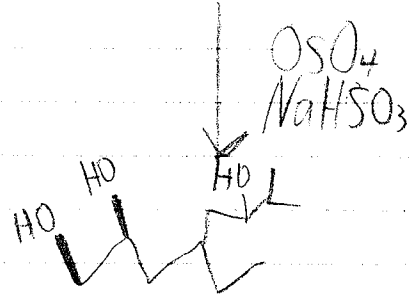
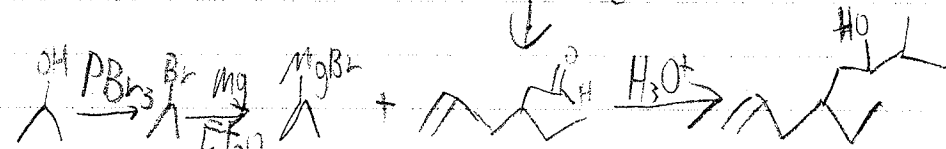
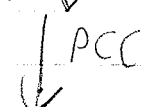
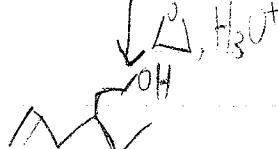
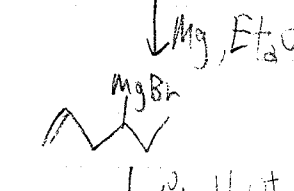
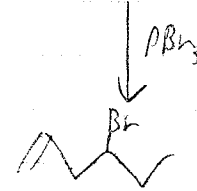
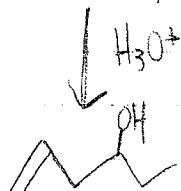


← not same as product at page bottom

Starting materials of 3 carbons or less



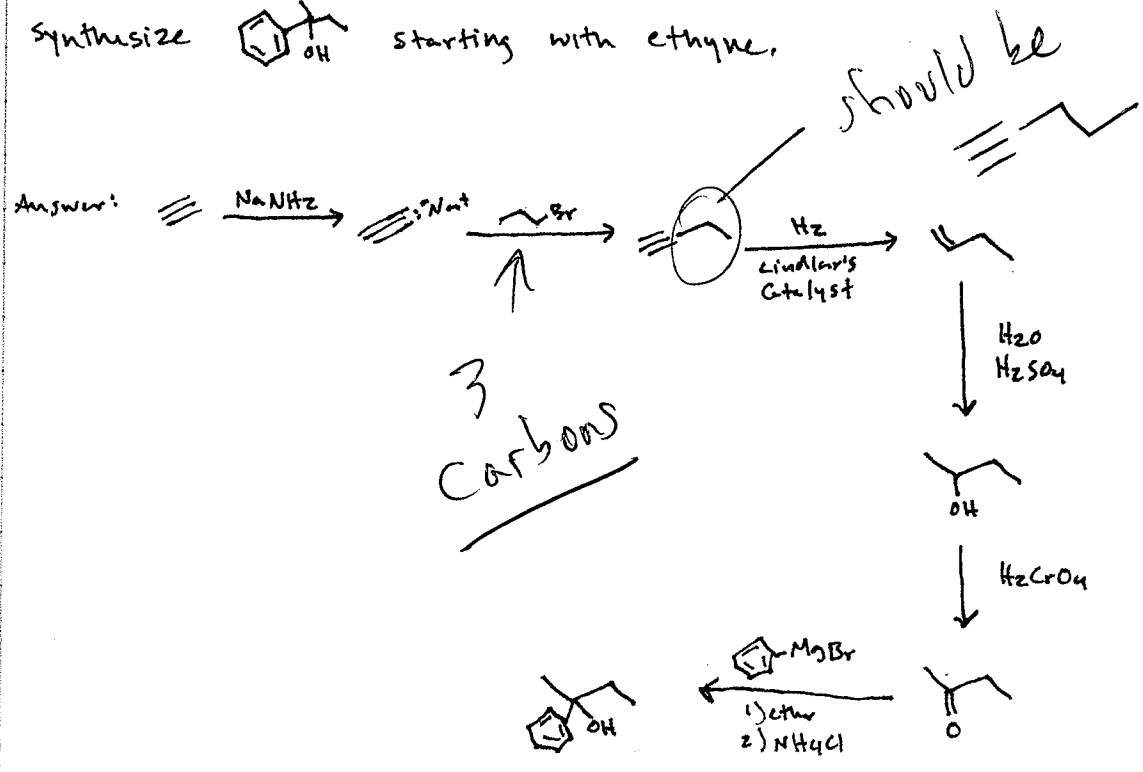
not good
~~WUWUWU~~
Meth



Homework: Synthesis Problem

Nina Miller
2/9/05

Synthesize C1=CC=CC=C1C(O)C starting with ethyne.

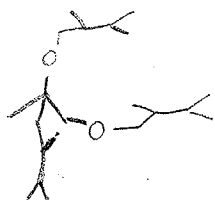


good
just 1 silly
error

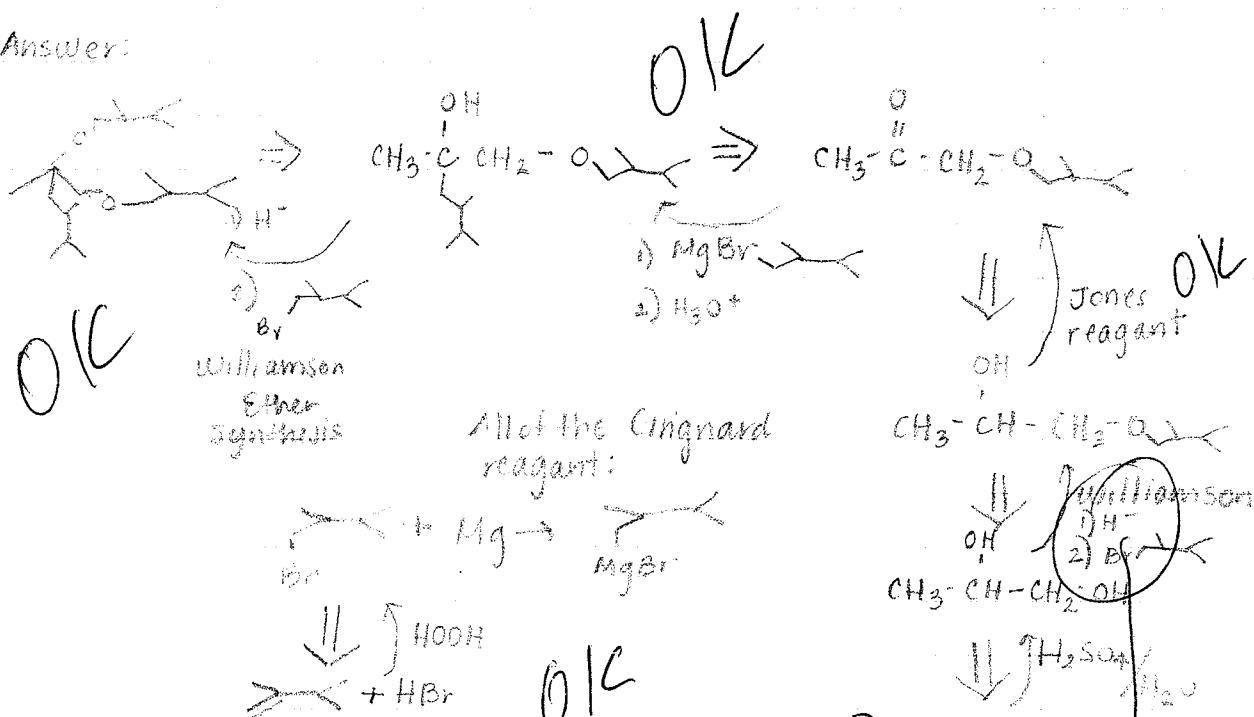
MB

Lrah Hampson
 Chem 336
 2/8/05

Synthesize the following: (2 carbon compound)

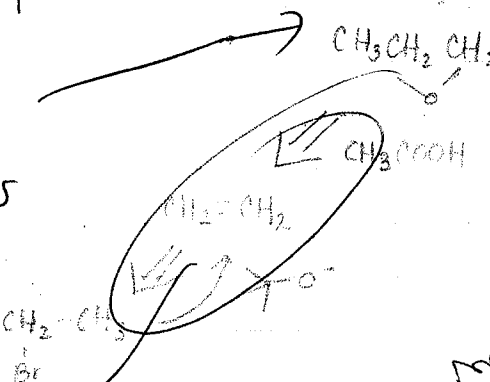


Answers:



I don't follow this one?

I don't follow?



You would make ethers out of both OH groups.